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**ENERGY**



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# Is Sodium the New Lithium? How Table Salt Might Save the Energy Storage Industry

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November 7, 2024

A Presentation of the Energy Storage Technology Advancement Partnership (ESTAP)

# Webinar Logistics

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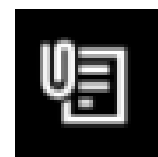
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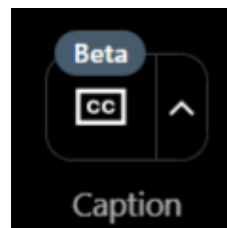
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**Speaker bios** available in the “Materials” section



Automated **captions** are available



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The Clean Energy States Alliance (CESA) is a national, nonprofit coalition of public agencies and organizations working together to advance clean energy.

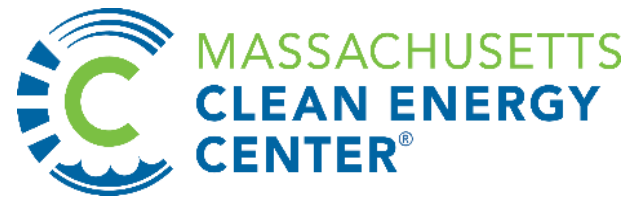
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# Energy Storage Technology Advancement Partnership (ESTAP)

Conducted under contract with Sandia National Laboratories, with funding from US DOE Office of Electricity.

- Facilitate public/private partnerships to support joint federal/state energy storage demonstration project deployment
- Support state energy storage efforts with technical, policy and program assistance
- Disseminate information to stakeholders through webinars, reports, case studies and conference presentations

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# Thank You!



## **Dr. Imre Gyuk**

Director, Energy Storage Research,  
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## **Waylon Clark**

Energy Storage Program Demonstration Team Lead,  
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# Upcoming Webinars

Demand Management: A Cost-Effective, Emissions-Free Alternative to New York City's Aging Peaker Power Plants (11/12)

Expanding Clean Energy Access and Benefits: Award-Winning Programs in Connecticut and Maryland (11/14)

Residential Solar+Storage: Weighing the Benefits of Bill Savings vs. Backup Power (11/20)

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# Decarbonization Needs Diversification!

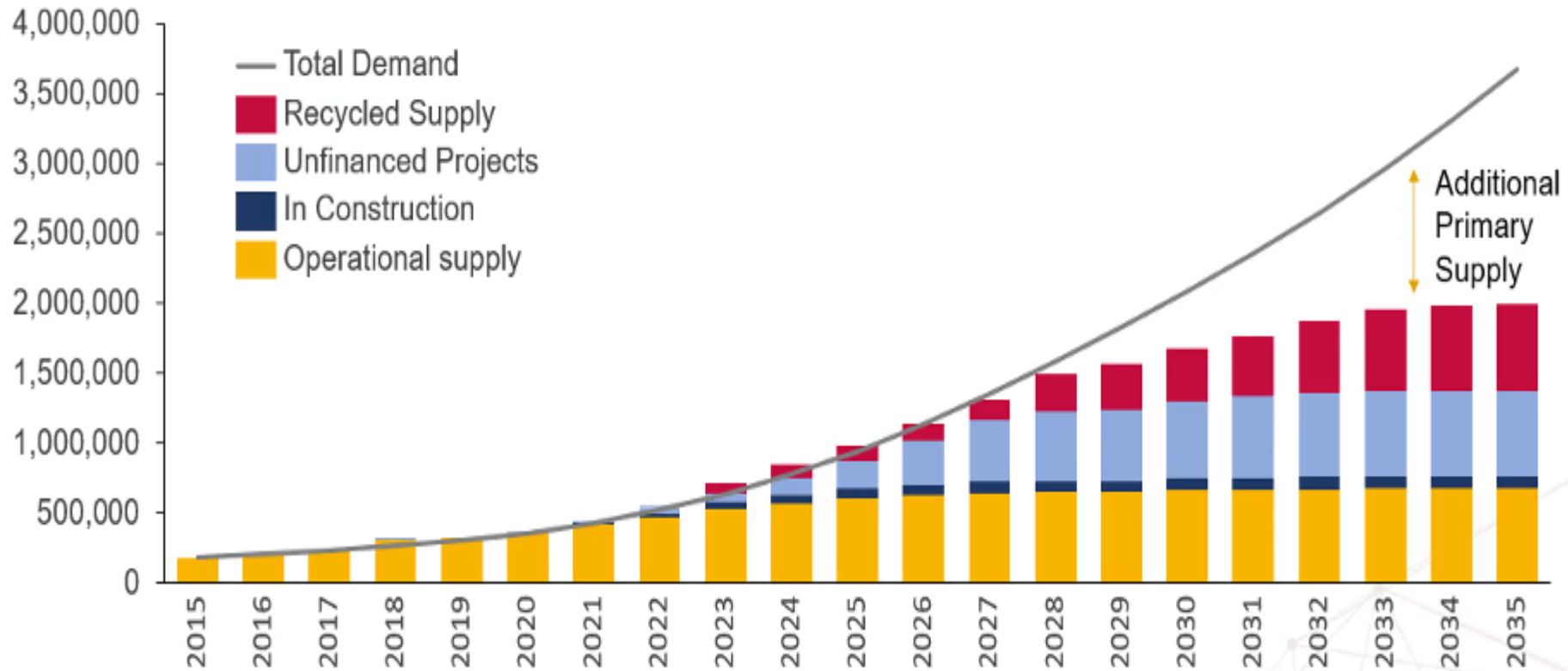
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IMRE GYUK, CHIEF SCIENTIST  
ENERGY STORAGE RESEARCH, DOE-OE

While Li is excellent for  
Short Duration Applications  
< 4 hours,  
Li is not expedient for  
Medium and Long Duration  
Energy Storage.

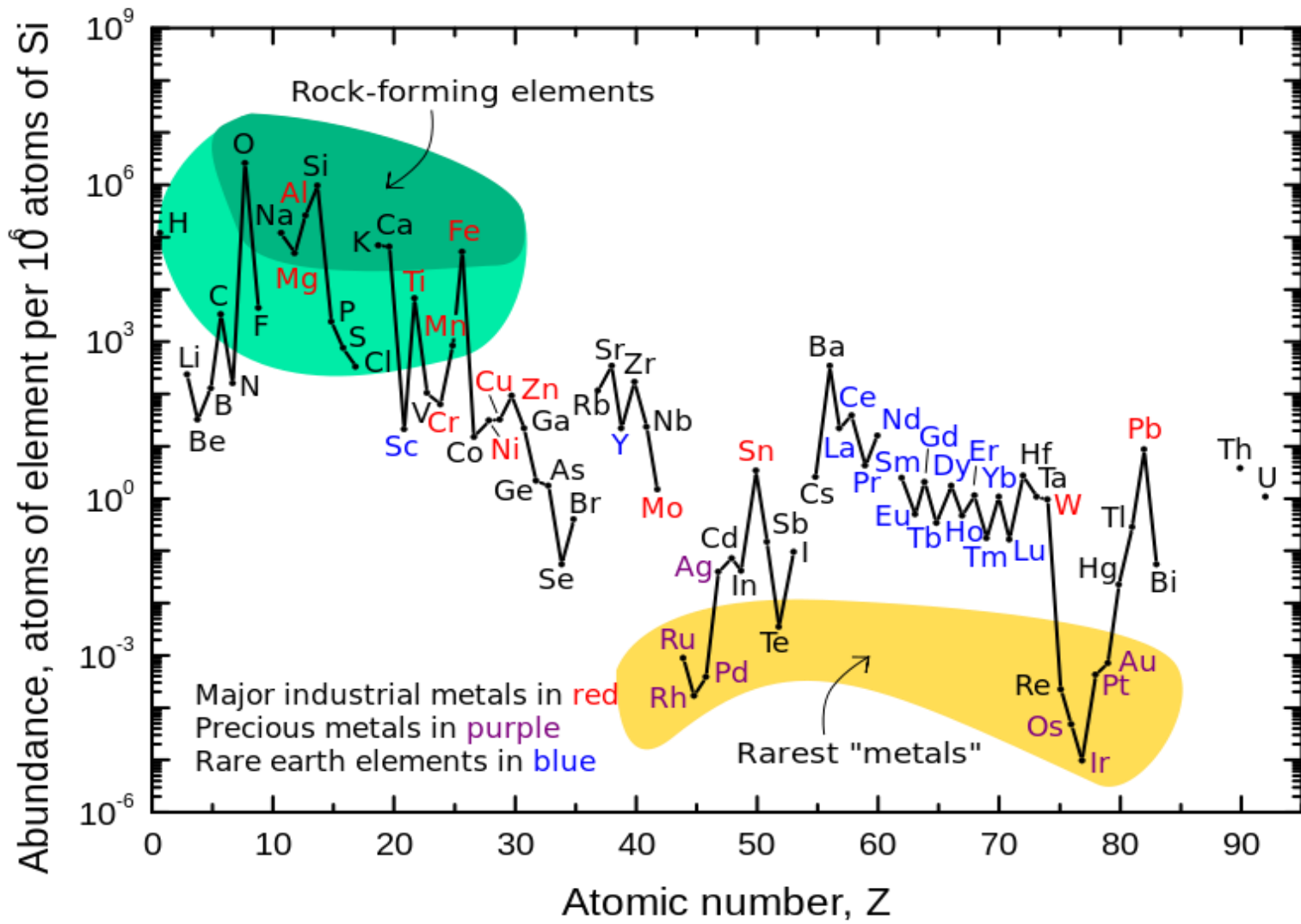
Furthermore, available Lithium  
will be needed for  
Vehicle Electrification.  
EVs require  
the high Energy Density of Li.

## Lithium Demand Vs Financed and Unfinanced Supply (MT LCE)



By 2029 Demand will outstrip Supply!

We are looking for Batteries  
using Earth-abundant,  
Domestically available,  
Inexpensive Materials,  
while showing  
Performance Characteristics  
Similar to Li-ion

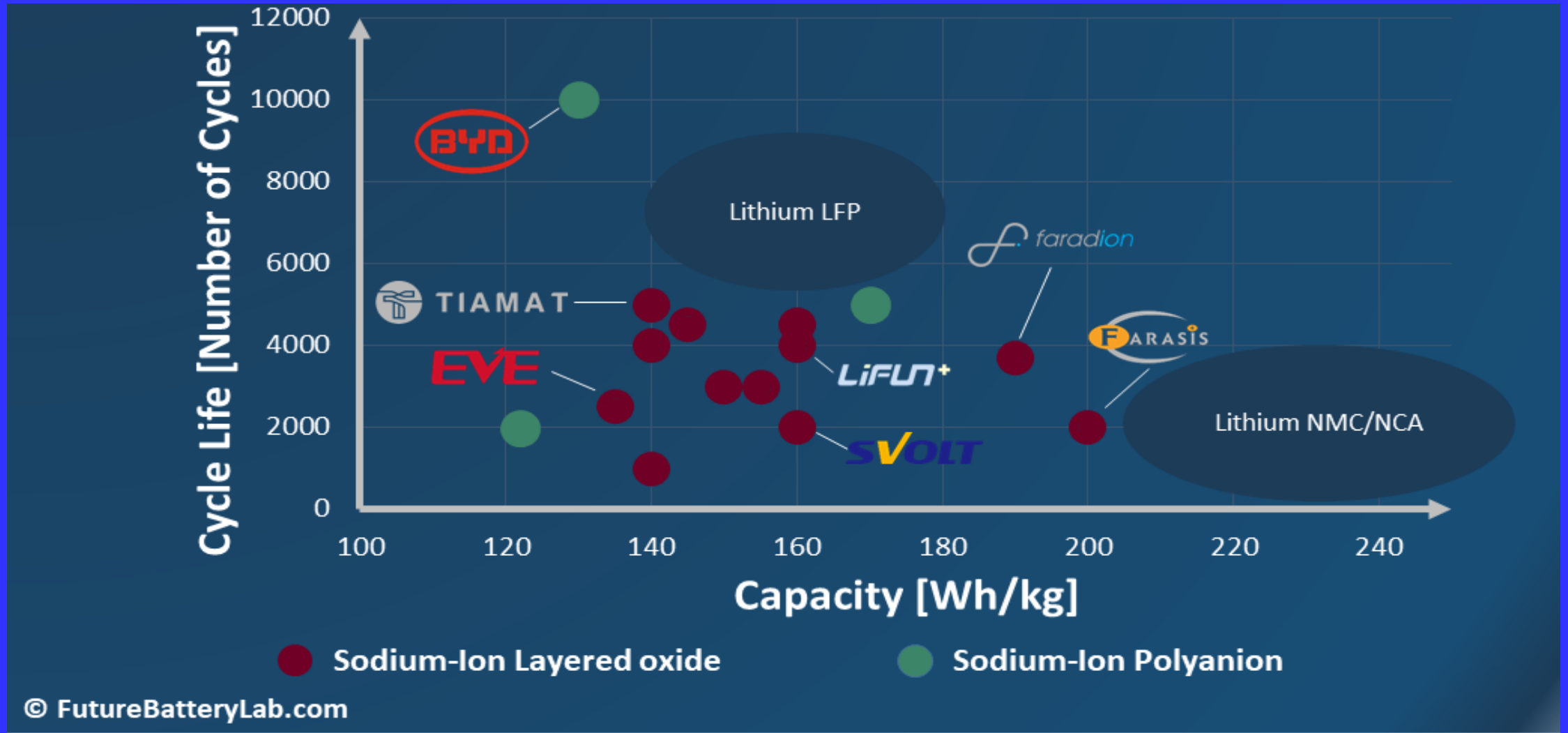


We want low Cost !

Li Carbonate: \$20,000 /ton

Na Carbonate: \$332 /ton

There are 135 Billion tons of  
minable Sodium Carbonate  
available in the Green River  
Basin of Wyoming



CYCLE LIFE vs CAPACITY for COMMERCIAL Na-ion BATTERIES



# GIGAFACTORIES

US: Natron

China: CATL, BYD

India: Reliance (Faradion)

Sweden: Altris



JAC EV powered by 25kWh Na-ion Battery – 250 km/charge

“BYD begins construction of 30GWh sodium-ion battery plant in China”



BYD, Na-ion, 30 kW / 100kWh

Na-ion Batteries lag Li-ion  
with respect to performance,  
but they are  
competitive on price and  
can be sourced domestically.  
They will find their market  
when Li becomes scarce.



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# Sodium Batteries – Liquid Sodium and Sodium Ion Batteries.



Ramesh Koripella, Ph.D.

CESA webinar presentation, Nov 07, 2024.



*This material is based upon work supported by the U.S. Department of Energy, Office of Electricity (OE), Energy Storage Division.*



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# Sodium ion batteries

- There is a lot of excitement about Na-ion batteries.
  - They are made with earth abundant materials that are lower in cost and safe.
  - No significant thermal runaway problems.
  - They can be shipped in zero charge state. - much safer during shipping.
- There are many different types of sodium batteries – low temperature, high temperature, different chemistries.
  - What are the differences?
  - How do they compare with Li-ion batteries?
  - What is the current state of commercialization?

# Outline



- Types of Sodium (Na) batteries
  - High temperature molten Na ion battery types
  - Low temperature Na ion batteries analogous to Li- ion batteries.
- Battery chemistries
- Pros and Cons, typical issues
- Current development status

*Information presented here is gathered mostly from the literature articles, publicly available information from conference presentations with some insights based on our discussions with SNL battery experts.*

# Types of Na ion batteries



## Types of Na ion batteries

1. **High temperature molten Na battery**
  2. **Low temperature Na ion battery, analogous to Li ion battery.**
- 
1. **Molten Na batteries** (high temperature batteries – few are commercial, few at R&D stage)
    - i. Na-S (Sodium – Sulphur) batteries – Commercial
    - ii. Na-Metal halide (ZEBRA) molten salt batteries – Commercial
    - iii. Intermediate and Low temperature molten Na batteries– still in R&D
  2. **Na-ion batteries** (low temperature analogues to Li ion batteries) – early commercial to precommercial stage.  
Different types based on differences in the choice of anodes and cathodes:
    - i. Layered Oxide
    - ii. Polyanion
    - iii. Prussian Blue Analogs

# Moten Sodium Batteries - Sodium-Sulphur Batteries

- **Na-S batteries:** One of the oldest battery chemistries, developed in 1960's.
- Anode is molten sodium (MP: 98 °C) , Cathode is molten sulphur (MP: 118 °C). Electrolyte is solid Beta alumina ceramic tube. It uses low cost raw materials, but operating temperatures are high.
- Discharge:  $xS + 2Na = Na_2S_x$  ( $3 \leq x \leq 5$ ), cell  $V_{oc} \sim 2.08$  V at  $>300^\circ C$

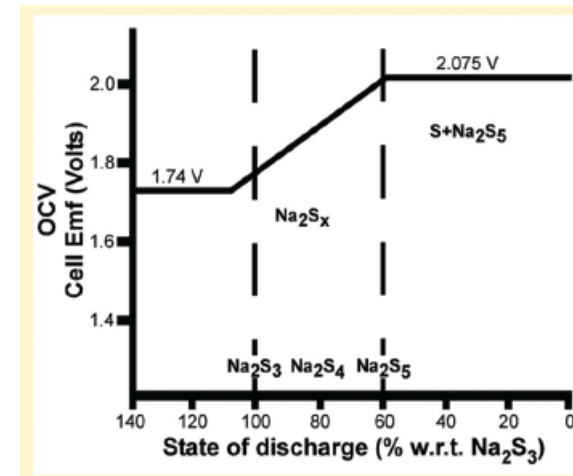
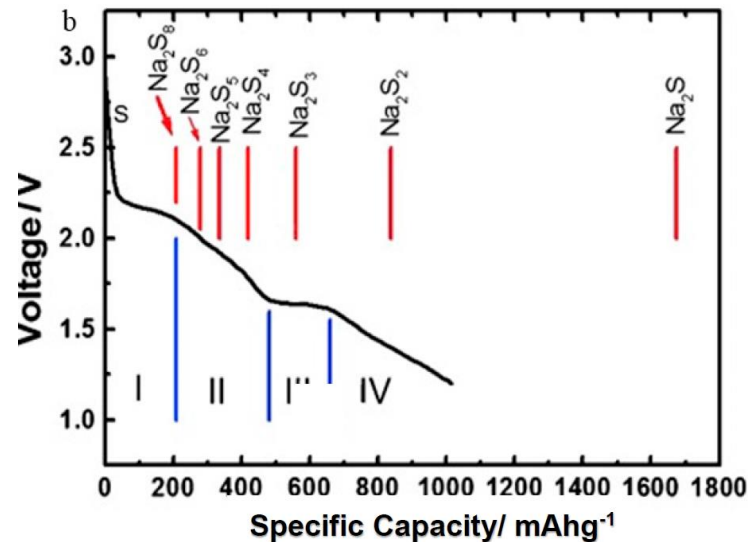
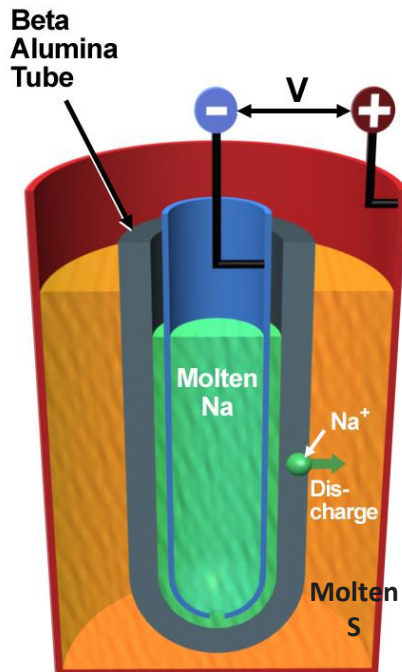


Illustration of a tubular molten Na-S battery configuration using a  $\beta''$ -alumina solid electrolyte (BASE).

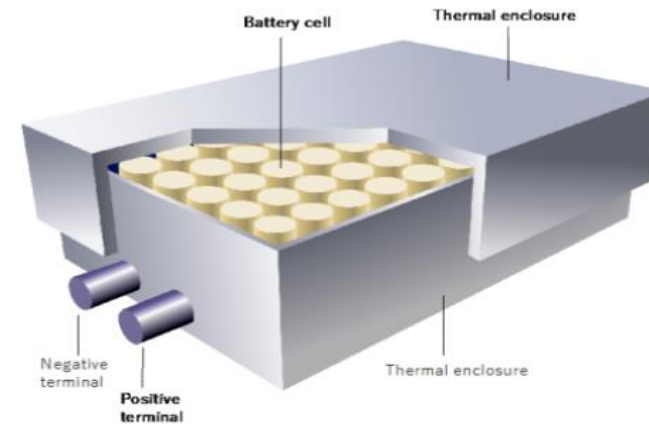
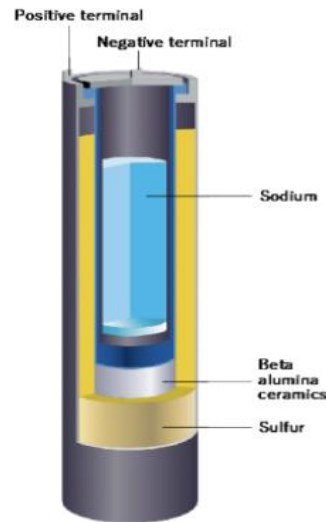
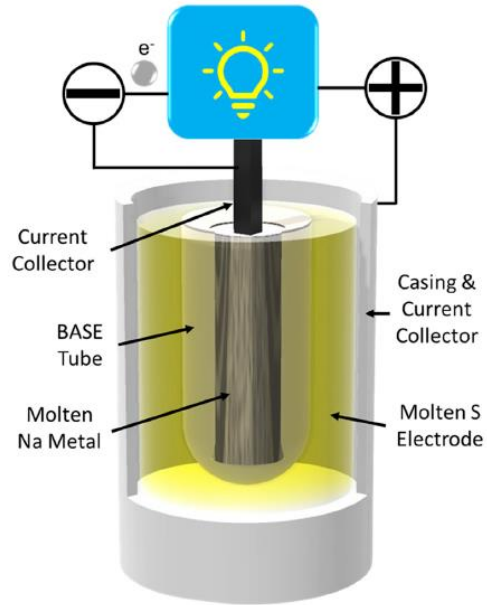
Developed in 1960s at Ford Motor Company.

- Typically fabricated in tubular form for ease of handling molten liquid metals. Difficulty with seals for planar configuration.
- High temperature operation ( $>300C$ ), needed for good ionic conductivity, but raises safety concerns.
- Theoretical specific capacity of Na-S is very high 1672 mAh/g, but due to the 1.74V discharge limitation, it drops to 557 mAh/g



# Commercial Sodium-Sulphur Batteries (NGK)

- Originally developed by Ford Motor Co in 1960's
- Commercialized by NGK for stationary power. 700MW/4.9GWh deployed mostly in Japan.



## Issues:

- High temperature (**300-350 °C**) **operation**. Thermal management is important, approximately **15%** daily **thermal loss** is normal. Need external heaters for start up. Targeted for 6-8 hr storage duration, energy density ~ 115 Wh/Kg, >5000 Cycle life, ~75% RTE and 15 yrs life.
- **Safety issues:** Handling two molten liquid electrodes Na (MP: 98 °C) and Sulphur (MP: 118 °C) separated by a Beta alumina solid electrolyte (BASE). Any cracks or failures in the ceramic Beta alumina electrolyte can cause short circuits by mixing the liquid Na and S, creating violent reactions and explosions.
- Metallic Na is combustible if exposed to moisture. **Cells are hermetically sealed, surrounded by sand in an insulated box.**
- **After one major safety incident, several design and operational improvements were made** with BMS controls with a good track record.

# BASF- NGK collaborative efforts on Na-S batteries

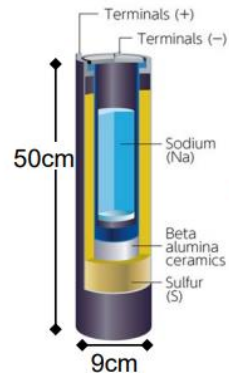


## NAS<sup>®</sup> Battery system design

### NEW PRODCUT RELEASE 2024

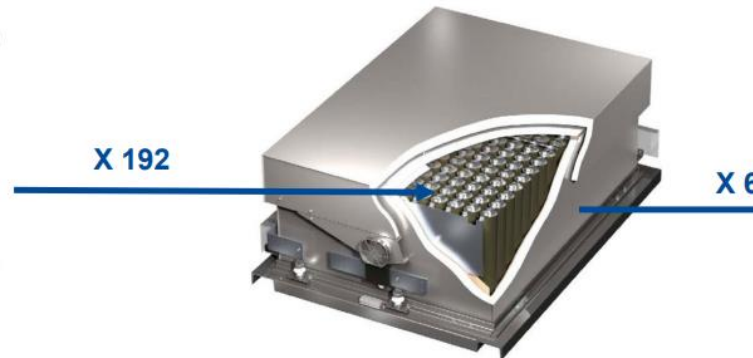
- Significantly improved heat management system
- Superior capacity degradation rates over lifetime

Battery cell



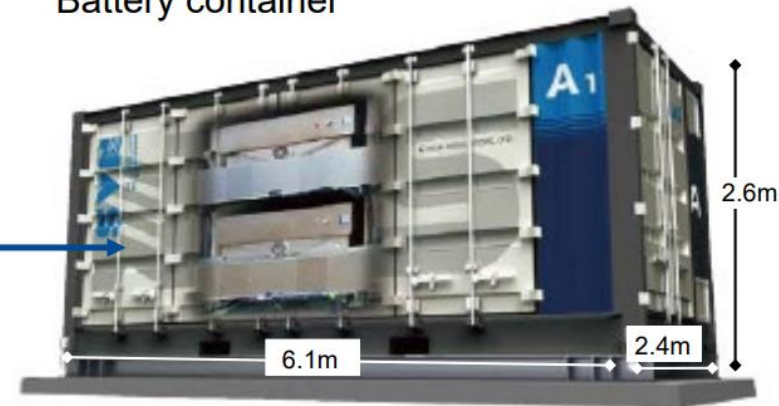
- 1.2kWh / 5.3kg
- ~2V
- C-rate 1/6 (0.17)
- T-range: 290°C – 360°C
- Life time: 7300 cycles or 20 years

Battery module



- 40kW / 245kWh

Battery container



- 20 feet container / 21 ton
- 250kW / 1.45MWh

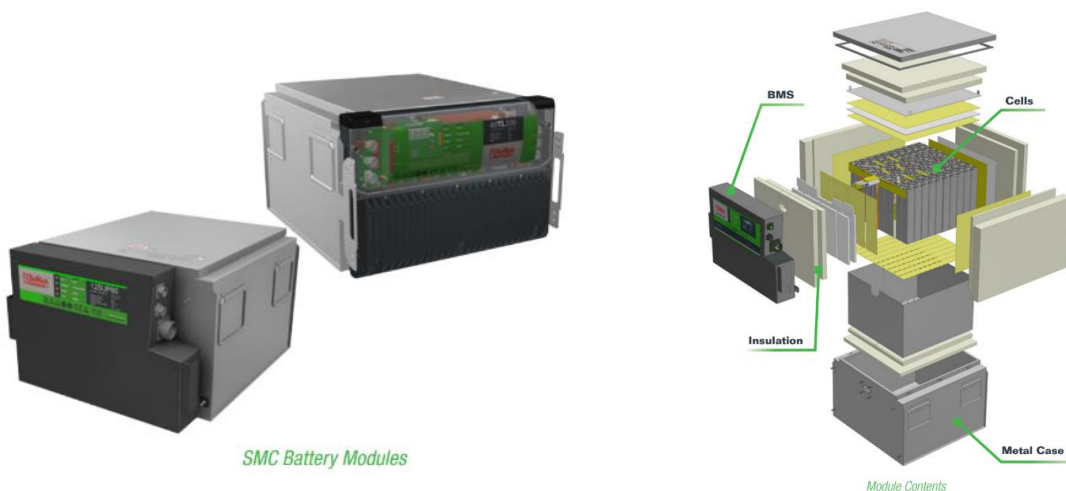
**BASF**  
We create chemistry

Several safety features were introduced. Successfully completed the UL 9540a, container level thermal runaway tests.

## Sodium Metal Chloride Batteries also known as ZEBRA batteries.



- Sodium metal chloride batteries also known as sodium metal halide batteries or ZEBRA batteries: Anode is molten Na, Cathode is molten sodium tetrachloroaluminate ( $\text{NaAlCl}_4$ ) salt mixed with  $\text{NiCl}_2$  (and some times  $\text{FeCl}_2$ ), electrolyte is modified Beta alumina ceramic. Low cost raw materials, high temperature operation.
- Discharge:  $\text{NiCl}_2 + 2\text{Na} = \text{Ni} + 2\text{NaCl}$ , cell  $V \sim 2.58 \text{ V}$  at  $250^\circ\text{C}$
- Currently being developed and commercialized by SoNick® systems, based in Switzerland. (*Similar Na metal chloride battery system was developed earlier by GE for locomotive applications and abandoned*)
- For the ZEBRA system, power density is  $\sim 166 \text{ Wh/kg}$ . Na- $\text{NaCl}_2$  operates at a lower temperature and slightly safer compared to Na-S batteries, but the higher cost of Ni is a challenge.



**FZSONICK ENERGY BACKUP**



# Other molten Sodium batteries – Adena Power

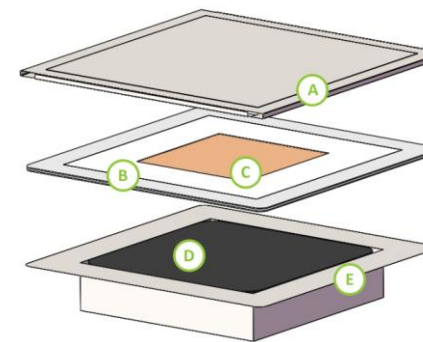


## Sodium Solid Electrolyte batteries for stationary energy storage

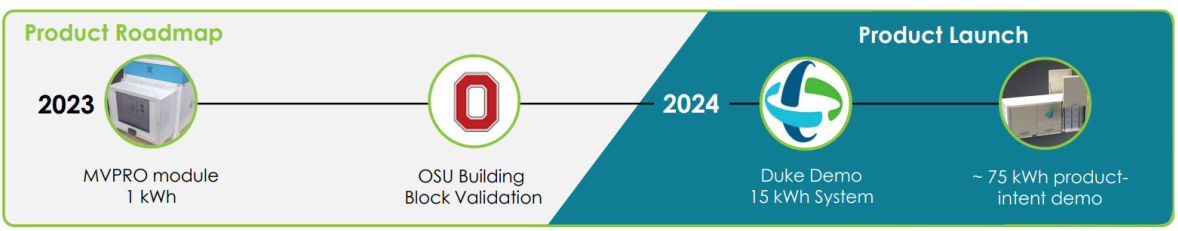
Demonstrations with utility partners to support C&I buying decision



## Adena's Cell Technology



- (A) Anode
  - o Formed in-situ
- (B) Sealing
  - o Simple polymer gasket
- (C) Polymer-supported NaSICON electrolyte
  - o Thin, conductive substrate
  - o Mechanically robust
- (D) Iron/salt cathode
  - o Capacity controlled by depth
- (E) Cell-housing
  - o Low-cost and scalable



- It utilizes **NaSICON** a slightly lower temperature (~125-150°C) electrolyte.
- High temperature plastic seals can be used.
- **Precommercial stage.**

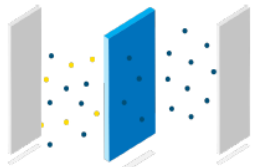


# Sodium-polysulphide flow Battery

## Enlighten's Energy Storage Technology

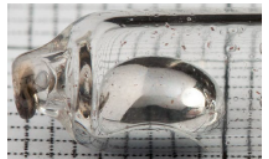


**NaSICON**




Solid State Na<sup>+</sup> Conductor for < 125 °C Operation

**Liquid Sodium Anode**



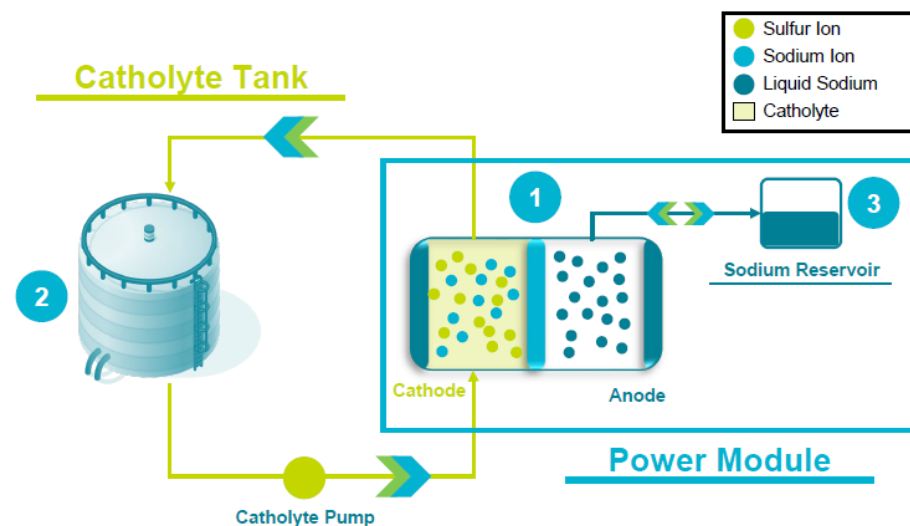
High Energy Density Liquid Metal for < 125 °C Operation

**Sodium Polysulfide Rich Catholyte**



High volumetric energy density, and low-cost

Three technical breakthroughs enable a low-cost, high-energy density, and perfectly decoupled energy storage solution based on sodium polysulfide flow battery chemistry



R&D stage.

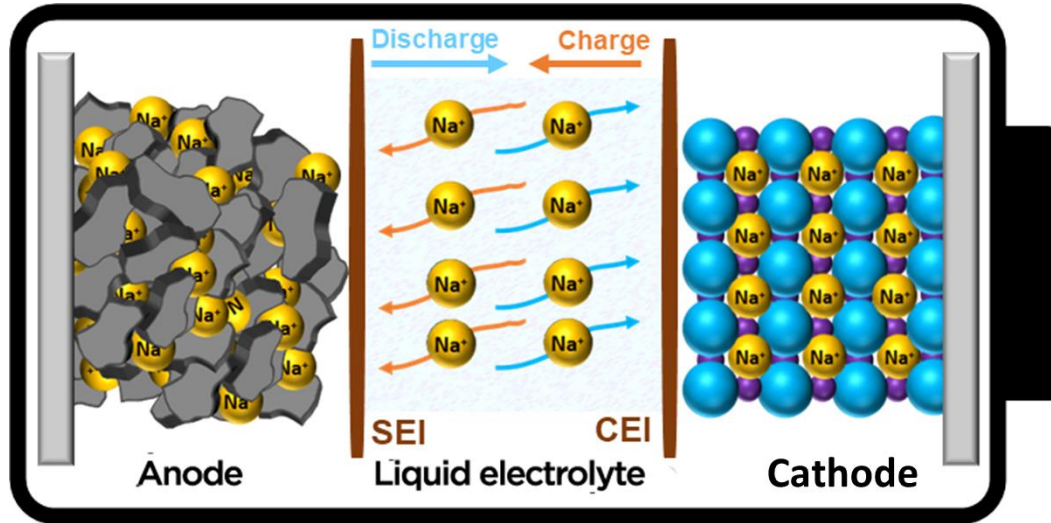


Switching gears to low temperature Na-ion  
batteries.

# Na-ion batteries



Na-ion batteries (SIB) are very similar to Li-ion batteries (LIB).



Na ions shuttle between the anode and cathode during charge and discharge very similar to Li ion battery.

Cathodes: Na based material of different types

Anodes: Hard Carbon, intercalation oxides, conversion & alloying

Electrolyte: mostly organic, few aqueous electrolytes

Separator: Polymer membrane very similar to Li ion battery

Current collectors: Al on both sides

- During charging Na ions move from the cathode to anode and the reverse process occurs during discharge.
- In Li ion batteries, Li ions shuttle between graphite anode and layered perovskite type cathode during charge and discharge. Graphite anodes (also Si anodes) used in Li-ion batteries are not a good choice for Na ion batteries.
- Through R&D efforts, suitable stable anode and cathode materials were developed for Na ion batteries and some versions are being commercialized.
- Several Chinese companies such as HiNa, CATL and Farasis announced mass production plans for commercialization. In the Europe, Faradion, Northvolt, Tiamat and in the US, Natron Energy have also announced commercialization plans.

# Sodium ion Battery Performance Comparison



	Lead acid	NiMH	Na-Ion	LFP	NMC
Energy density (Wh/Kg)	30-50	60-120	70-160	90-160	200-250
Nominal Voltage (V)	2	1.25	3.1-3.2	3.2	3.6
Cycle life (80% retention)	200-300	300-1000	>1000	5000	1000-2000
Self discharge [month]	5-10%	15-20%	5%	2-3%	0.5-2.5%
Est. Cost (\$/kWh)	50-100	250	50-80	60-95	90-110
Operating Temp. range (°C)	-20C to 70C	-40C to 50C	-40C to 80C	-20C to 60C	-20C to 60C

- Performance of Na ion batteries are comparable to LFP batteries, their biggest competition.
- Even though Na is much less expensive than Li, the difference are only in the cathode materials and in the current collectors of both batteries, rest of the materials and processing costs would be similar. May be 10-20% cost advantage.



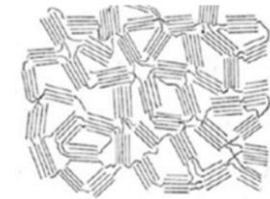
# Na-ion battery types and material choices



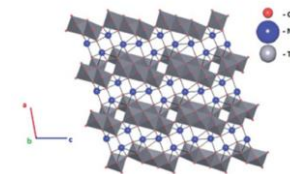
- There are different varieties of Na ion batteries depending on the choice of materials used for the anode, cathode and the electrolyte. Voltage output, performance, cycle life and cost depend on the materials choice.
- **Anode materials:** 3 types - Hard Carbon, Ti based intercalation oxides, conversion & alloying anodes.

- **Hard Carbon** – current standard anode. **Pros:** decent specific capacity (~300 mAh/g), voltage and rate performance. **Cons:** limited capacity, initial irreversible capacity loss, Na plating.
- **Ti based intercalation oxides:**  $\text{TiO}_2$ ,  $\text{Na}_2\text{Ti}_3\text{O}_7$  etc. **Pros:** low working potential (0.3V for  $\text{Na}_2\text{Ti}_3\text{O}_7$ ). **Cons:** low electronic conductivity and Na diffusion kinetics, low capacity.
- **Conversion and alloying:** Sn, Pb, P, Sb etc. **Pros:** High capacity (>500 mAh/g). **Cons:** large volume change, low electronic conductivity.

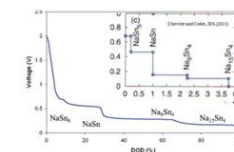
[Sodium is known to electrochemically and reversibly alloy with several group IV and V elements including phosphorus (P), germanium (Ge), tin (Sn), antimony (Sb), lead (Pb), and bismuth (Bi). Alloying materials store Na through a series of reactions involving bond breaking and forming, leading to the formation of the various alloy phases ( $\text{Na}_x\text{M}$ , where  $\text{M} = \text{P}, \text{Ge}, \text{Sn}, \text{Sb}, \text{Pb}, \text{or Bi}$ ).] ([UNIGRID](#))



$\text{TiO}_2$ ,  $\text{Na}_2\text{Ti}_3\text{O}_7$ , etc.



Sn, Pb, P, Sb, etc.



# Na-ion battery types and material choices



**Cathode materials:** different types such as; Layered oxides, Polyanionic, Prussian blue analogs.

- **Layered oxides:**  $\text{Na}_x\text{MO}_2$  (*Faradion*), capacity  $\sim 190$  mAh/g
- **Polyanionic:** NASICON, NaFP, NVP, NVPF (*Tiamat*), capacity  $\sim 117$  mAh/g
- **Prussian Blue:**  $\text{Na}_x\text{MFe}(\text{CN})_6$  and other variations (*Natron Energy*, *CATL* -Prussian white). capacity  $\sim 165$  mAh/g

**Electrolytes:** mostly organic solvents, Na-salts.

- Organic solvent: Ethylene carbonate (EC), propylene carbonate (PC)
- Na-salts:  $\text{NaClO}_4$  salts,  $\text{NaPF}_6$  salts, and other R&D efforts
- Aqueous Electrolytes: Natron uses aqueous electrolyte, results in lower cell voltage compared to organic electrolyte systems. Ex:  $\text{NaNO}_3$ ,  $\text{NaClO}_4$  with other additives.

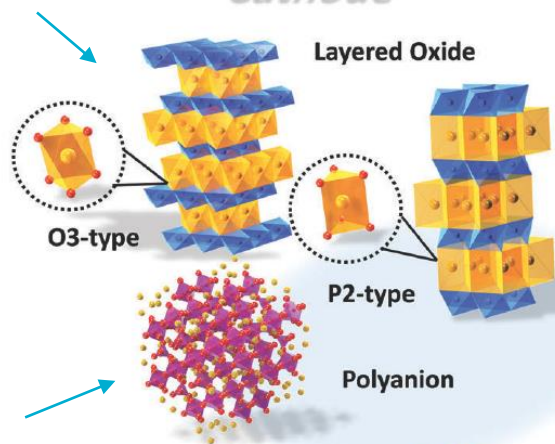
**Current Collectors:** Al metal is used at both cathode and anode (Lower cost, less inventory). (*For Li-ion batteries, Cu at the anode and Al at the cathode are used.*)

# Na - ion Battery Types Schematic

Ex: Transition metal oxide of type P3, P2 or O3.  $\text{Na}_{0.76}\text{Mn}_{0.5}\text{Ni}_{0.3}\text{Fe}_{0.1}\text{Mg}_{0.1}\text{O}_2$   
(Intercalation of Na)

*Faradion, NorthVolt*

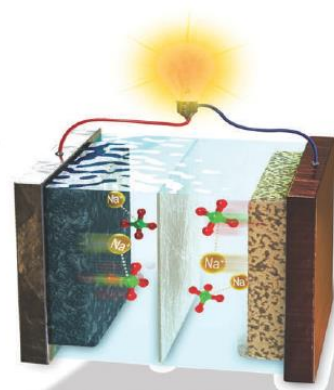
## Cathode



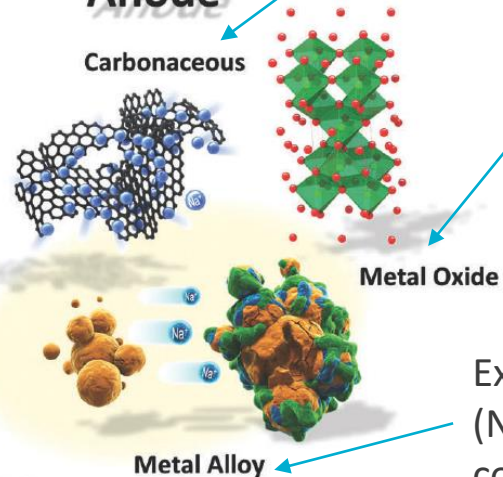
Ex: sodium - vanadium - phosphate - fluoride. Prussian blue analogues ( $\text{Na}_2\text{MnFe}(\text{CN})_6$ ).

*Natron Energy, CATL*

## Na-ion Battery



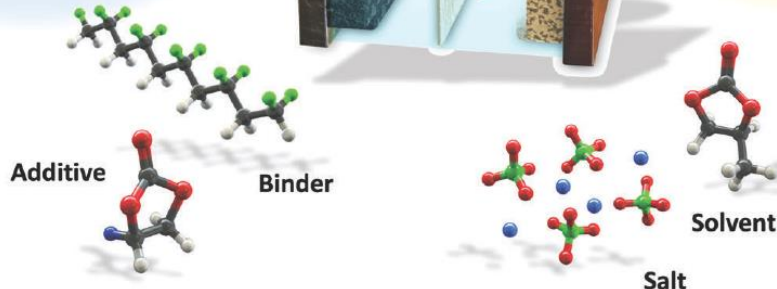
## Anode



Ex: Hard Carbon  
(Intercalation of Na)  
*Faradion*

Ex:  $\text{Na}_2\text{Ti}_3\text{O}_7$  or  $\text{NaTiO}_2$   
(Intercalation of Na)

Ex: Pb, P, Sn, Ge, etc.  
(Na alloying with metals or conversion reaction,  $\text{FeS}_2$ )  
*Unigrid*



## Electrolyte & Binder

Ex: typically non-aqueous electrolytes with  $\text{NaClO}_4$  and  $\text{NaPF}_6$  salts dissolved in EC, PC, DMC, DEC. Few aqueous types.

## Comparison of few Na ion batteries under development

	<b>Faradion</b>	<b>Tiamat</b>	<b>HiNa Battery</b>	<b>Natron Energy</b>
Anode	Hard Carbon	Hard Carbon	Hard Carbon	Prussian Blue
Cathode	Layered Oxide ( $\text{Na}_x\text{Ni}_{1-x-y-2}\text{Mg}_x\text{Mn}_y\text{Ti}_z\text{O}_2$ )	Polyanionic ( $\text{Na}_3\text{V}_2(\text{PO}_4)_2\text{F}_3$ )	Layered Oxide ( $\text{Na}_w\text{Cu}_x\text{Fe}_y\text{Mn}_z\text{O}_2$ )	Prussian Blue
Electrolyte	Organic Liquid	Organic Liquid	Organic Liquid	Aqueous Liquid
Voltage (V)	3.0 -3.15	~3.7	~3.2	~1.56
Cycles	1000-3000	4000	> 4500	35000
est. Energy density (Wh/Kg)	130-190	90-160	140-155	70

- Natron uses aqueous electrolyte, demonstrated long cycle life, but energy density is low.
- Other developers uses organic electrolyte with high energy density but cycle life is low.

# Sodium-ion batteries summary



- Compared to Li ion batteries, Na ion batteries are lower in cost, and uses earth abundant materials.
  - No Lithium, Cobalt and very low or no Nickel.
  - Utilizes Aluminum current collectors on both sides.
  - Main differences between the two are in cathode materials and current collectors.
  - Manufacturing processes are similar. Approx. 10-20% cost advantage over Li-ion batteries.
- Lower energy density than NMC batteries, but performance is comparable to LFP, their main competition.
- Offers better low temperature performance compared to Li ion batteries.
- They can be charged and discharged faster than Li ion batteries.
- Na ion batteries can be stored and transported in zero voltage. Safe for transportation. Li ion batteries should be stored at some minimum charge.
- The Na ions are larger than the Li ions and causes more stresses in the anodes and cathodes during intercalation leading to structural failures during charge/discharge cycles. Prussian Blue analogues have 3D structures with large channels for Na ion diffusion offering good cycle life.
- Main challenges are; increasing the cycle life, increasing the energy density and scaling up for high volume manufacturing.

Thank you for your attention

## Questions?

### Acknowledgements

This material is based upon work supported by the U.S. Department of Energy, Office of Electricity (OE), Energy Storage Division.





*Exceptional service in the national interest*

# MULTI-SCALE SAFETY EVALUATION OF COMMERCIAL SODIUM-ION CELLS AND MATERIALS

Alex M. Bates

*Sandia National Laboratories*

ESTAP Webinar, Sodium-Ion Batteries

November 5, 2024

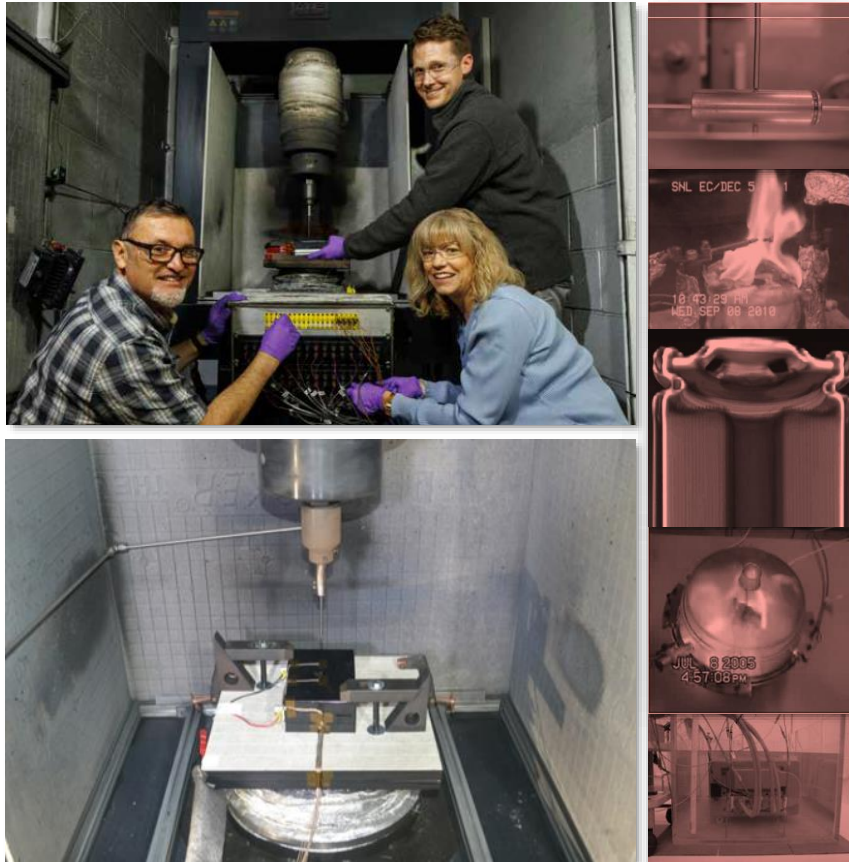


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# Sandia National Laboratories – Capabilities and Infrastructure



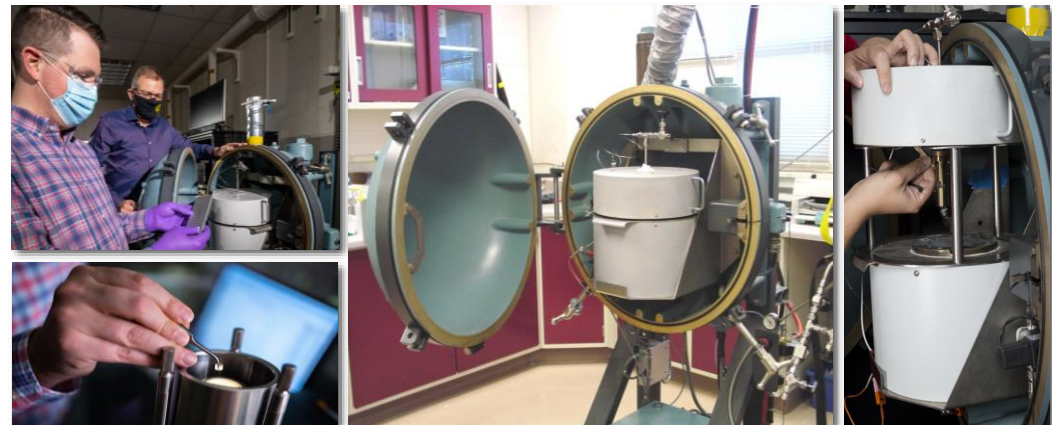
## Cell and Module Testing (<1kWh) Battery Abuse Testing Laboratory (BATLab)



## Battery Pack/System Testing Thermal Test Complex (TTC) and Burnsite (<100kWh)



## Battery Calorimetry (multi-scale)

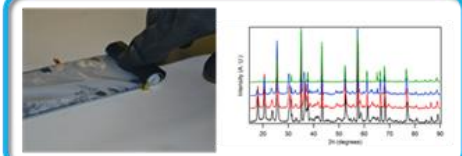




# BATLab Destructive Testing – Demonstration Reel



# Exploring All Aspects of Battery Safety and Reliability



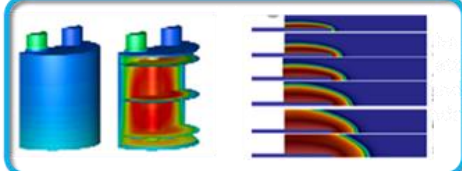
## Materials R&D

- Thermal stability and aging impact on battery components
- Vent gas composition
- Solid state battery safety
- Aqueous battery gas evolution



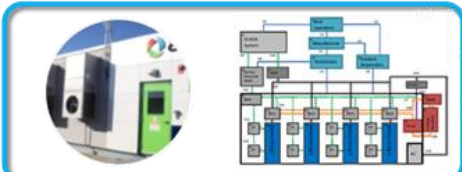
## Cell and Module Testing

- High precision cell cycling and degradation
- Electrical, thermal, mechanical abuse testing
- Failure propagation testing on batteries/systems
- Diagnostics during and post battery failure



## Simulations and Modeling

- Multi-scale models for understanding thermal runaway and failure propagation
- Fire Dynamic Simulations to predict the size, scope, and consequences of battery fires



## System Level Design and Analysis

- Hazard analysis methods to avoid fire and explosion
- Predictive maintenance
- Improved control using power electronics



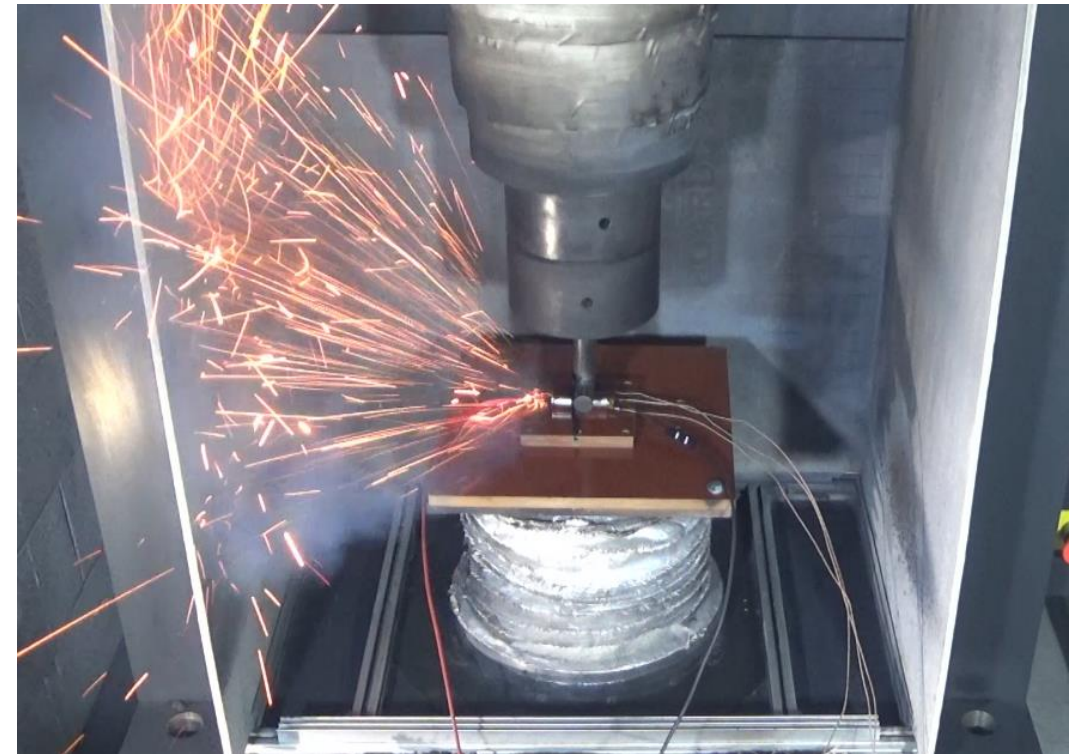
## Outreach, Codes, and Standards

- USABC Abuse Testing Manual (2022)
- Energy storage safety working group
- IEEE battery management system standard
- EPRI energy storage data submission guidelines

# Motivation for Na-ion Safety Testing



*Rechargeable, room temperature operation Na-ion technology has quickly transitioned to commercial availability with the promise of similar to Li-ion performance characteristics and fabrication requirements, and lower materials cost.*



Na-ion crush test performed at SNL.

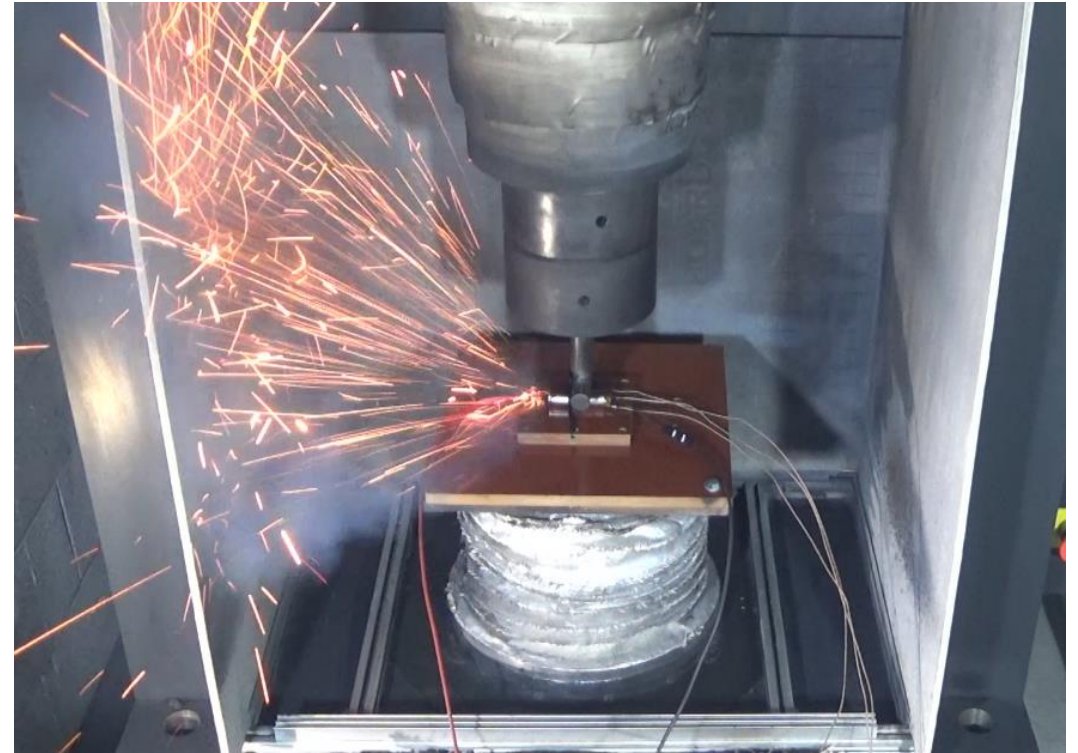
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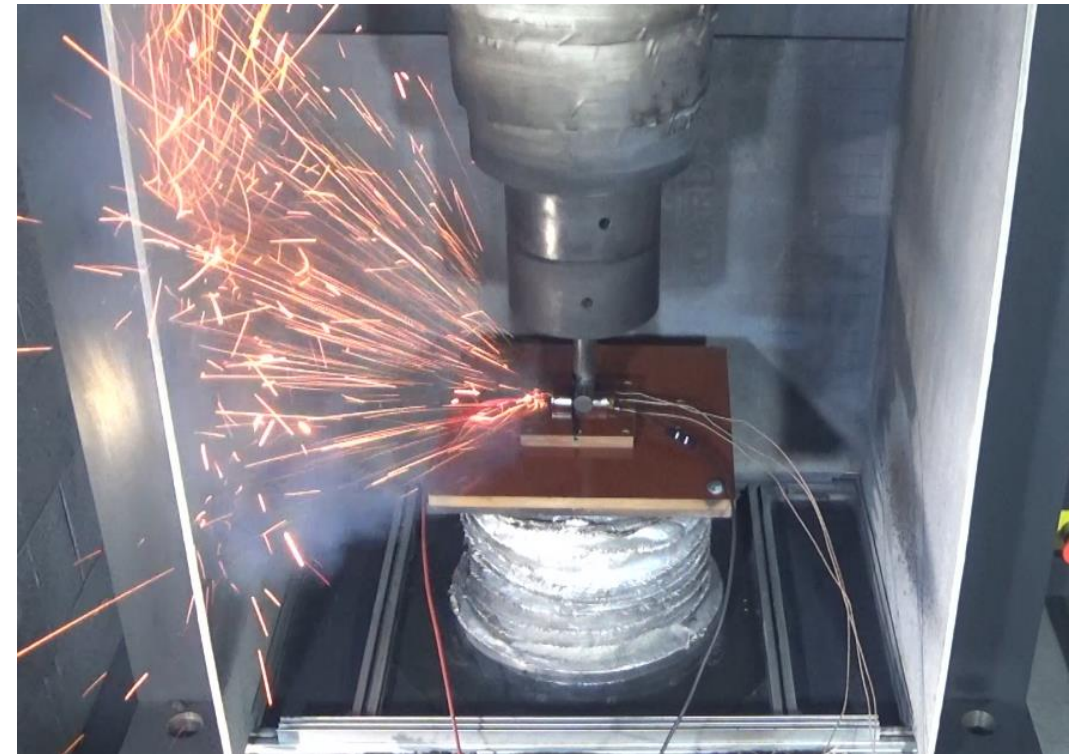
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## **This Presentation:**

1. Destructive Testing Comparison of:
  - 1.5 Ah Na-ion 18650 cell
  - 1.5 Ah Li-ion (NMC) 18650 cell
2. Introduction to Materials Scale Analysis



Na-ion crush test performed at SNL.

# Na-ion Grid Installations



<https://cnevpost.com/2024/05/13/china-1st-large-sodium-battery-energy-storage-station-operation/>

- Operational May 11, 2024
- 10 MWh
- Nanning, Guangxi, China
- China Southern Power Grid



<https://www.yicaglobal.com/news/worlds-largest-sodium-ion-battery-project-starts-operation-in-china>

- Second Phase
- 100 MWh



18650  
DESTRUCTIVE  
TESTING

# Na-ion Material Composition

## Na-ion 18650 cells purchased commercially

Specification	Value	Notes
Rated Capacity	1.5 Ah	123 Wh/kg (including case)
Experimental Capacity	1.43 Ah	1.5 to 4.1 V, C/5 rate, C/20 taper charge
Cathode	$\text{Na}_{2.5}\text{Ni}_1\text{Mn}_1\text{Fe}_1\text{O}_2$	ICP-MS
Anode	Hard Carbon	XRD
Electrolyte – Salt	$\text{NaPF}_6$	Assumed based on ICP-MS
Electrolyte – Solvent	46% PC, 21% EC, 14% EMC, 13% DMC	Area% from GCMS





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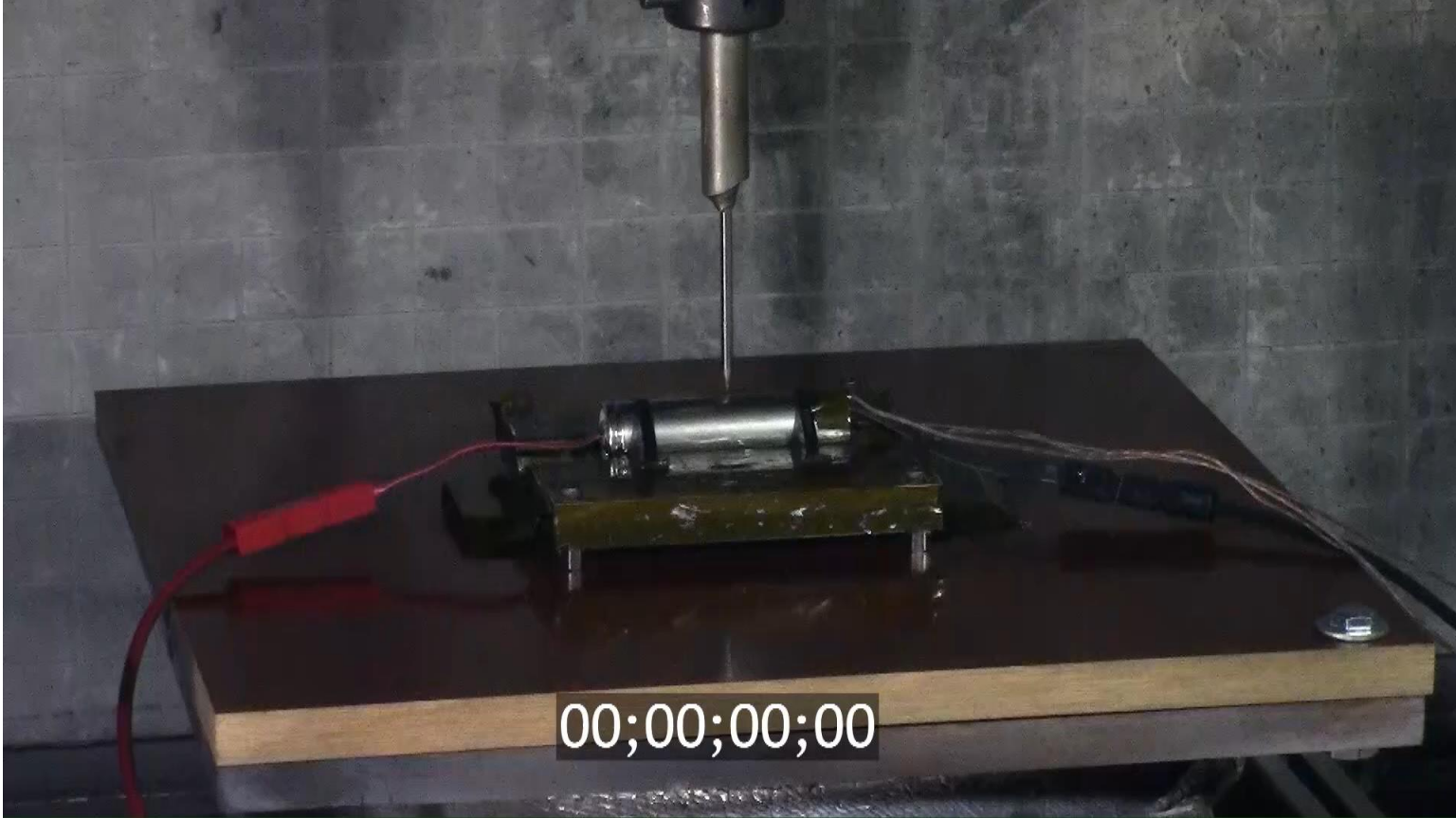
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Company	Cathode	Anode	Electrolyte
Natron Energy (US)	Prussian Blue Analog	Prussian Blue Analog	Aqueous
Novasis (US)	Prussian Blue Analog	Hard Carbon	Organic
Faradion (UK)	Layered Oxide	Hard Carbon	Organic
CATL (China)	Prussian Blue Analog	Hard Carbon	Organic
TIAMAT (France)	NVPF	Hard Carbon	Organic
Rechargion (India)	Olivine	Hard Carbon	Organic



# Nail Penetration, Na-ion Cell Close-up Video

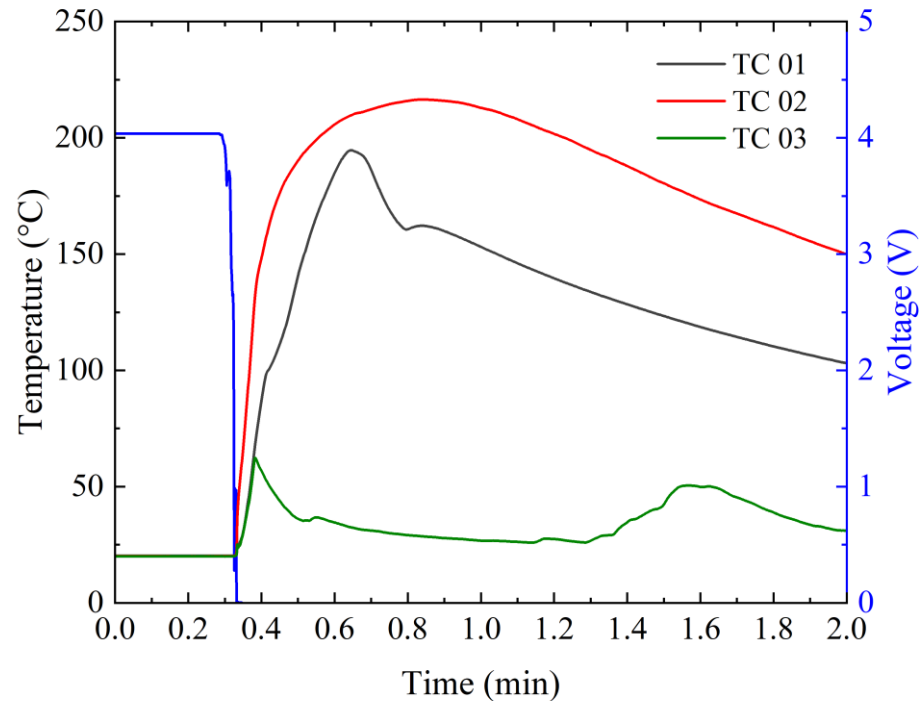


# Nail Penetration – Horizontal Orientation

## 100% SOC, 3 mm sharp nail, 2 mm/s

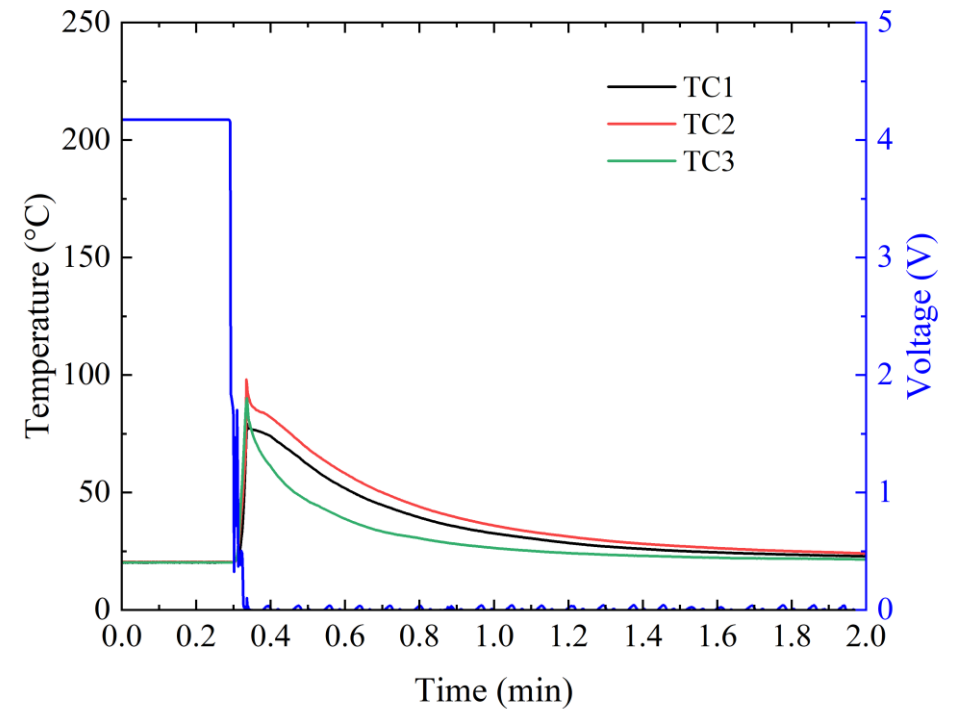


### Na-ion



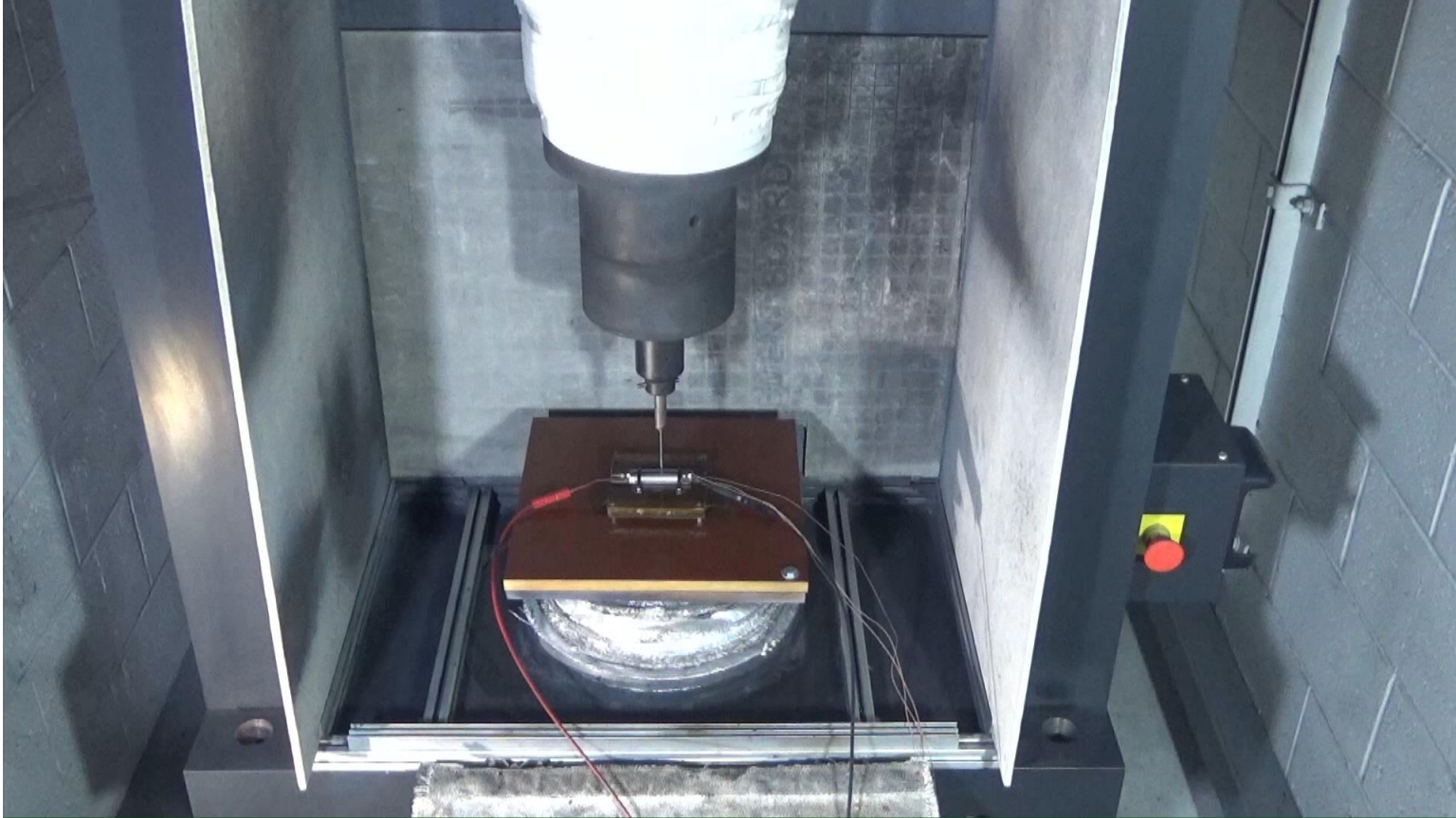
- Max temperature: 216.5 °C
- Nail penetration depth at voltage drop: 1.86 mm
- Venting, smoke, sparks, **violent pressure release**

### Li-ion

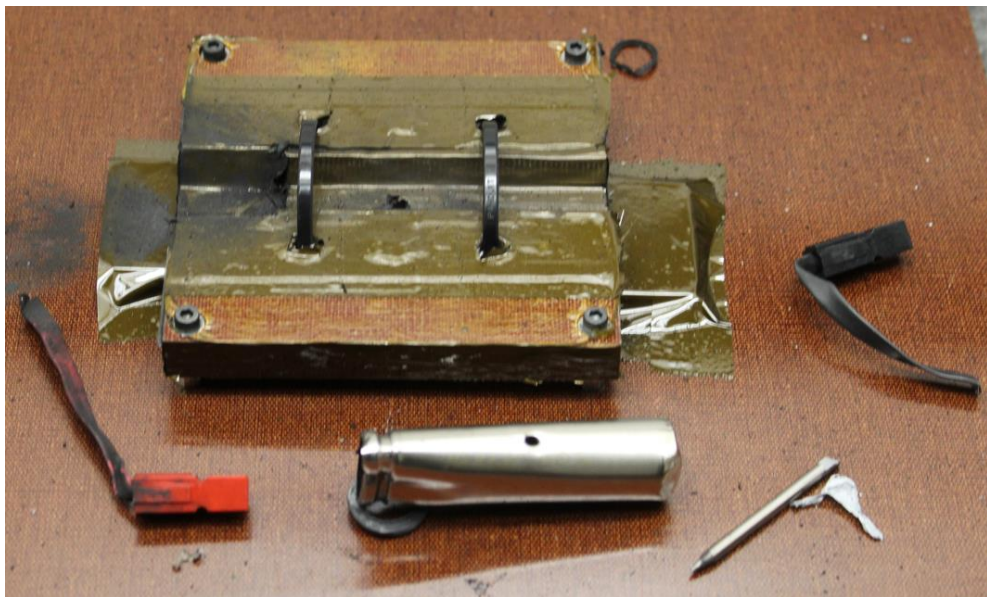


- Max temp: 98.1 °C
- Nail penetration depth at voltage drop: 1.84 mm
- No observable features

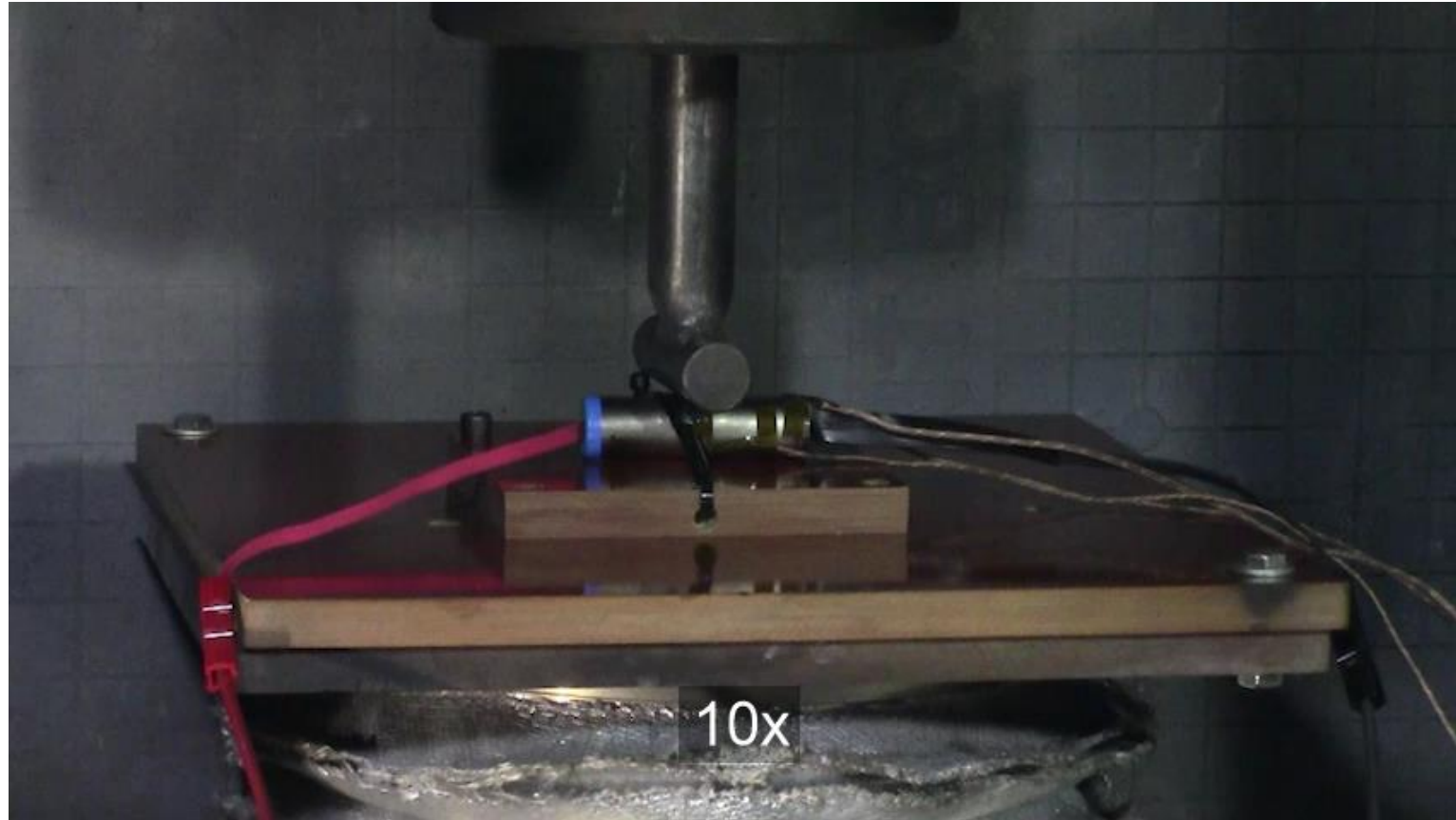
# Nail Penetration, Na-ion Cell Wide-angle Video



# Nail Penetration, Na-ion Cell Post-test Images

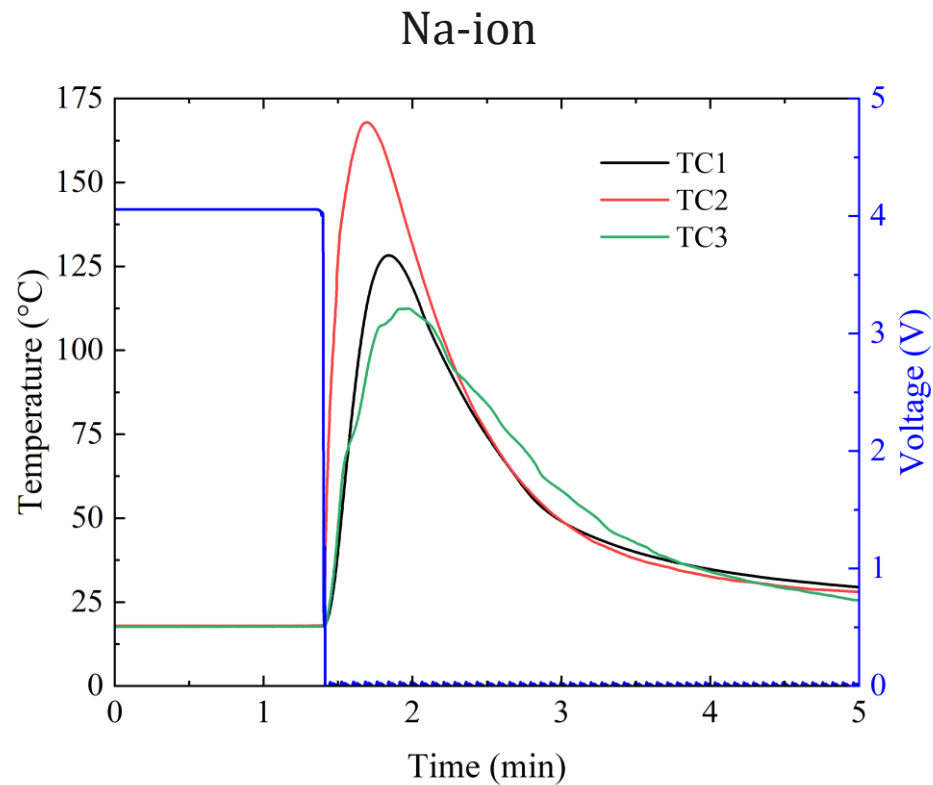


# Mechanical Crush, Na-ion Cell

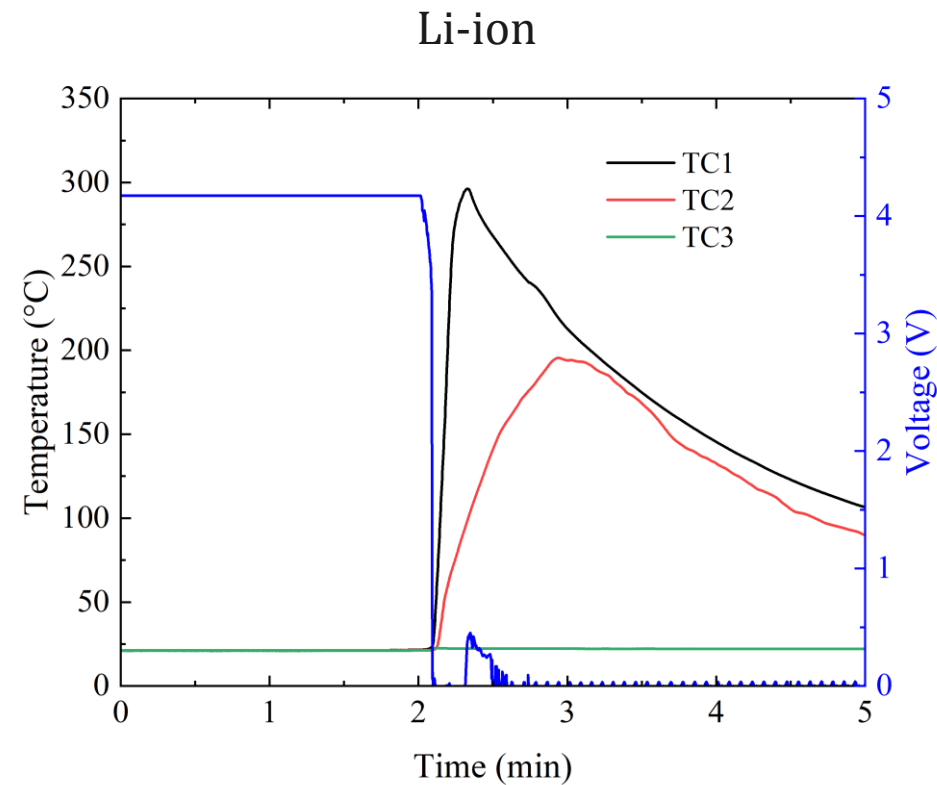


# Mechanical Crush

## 100% SOC, Hammer Implement, 0.1 mm/s

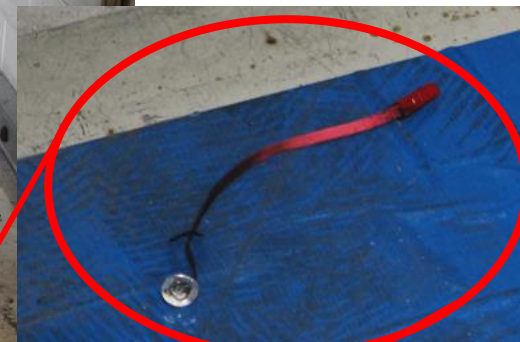


- Max temp: 167.9 °C
- Crush depth at voltage drop: 6.46 mm
- Vent, smoke, sparks, lid pop, flame



- Max temp: 296.3 °C
- Crush depth at voltage drop: 9.61 mm
- Vent, smoke, sparks, lid pop

# Mechanical Crush, Na-ion Cell Post-test Images



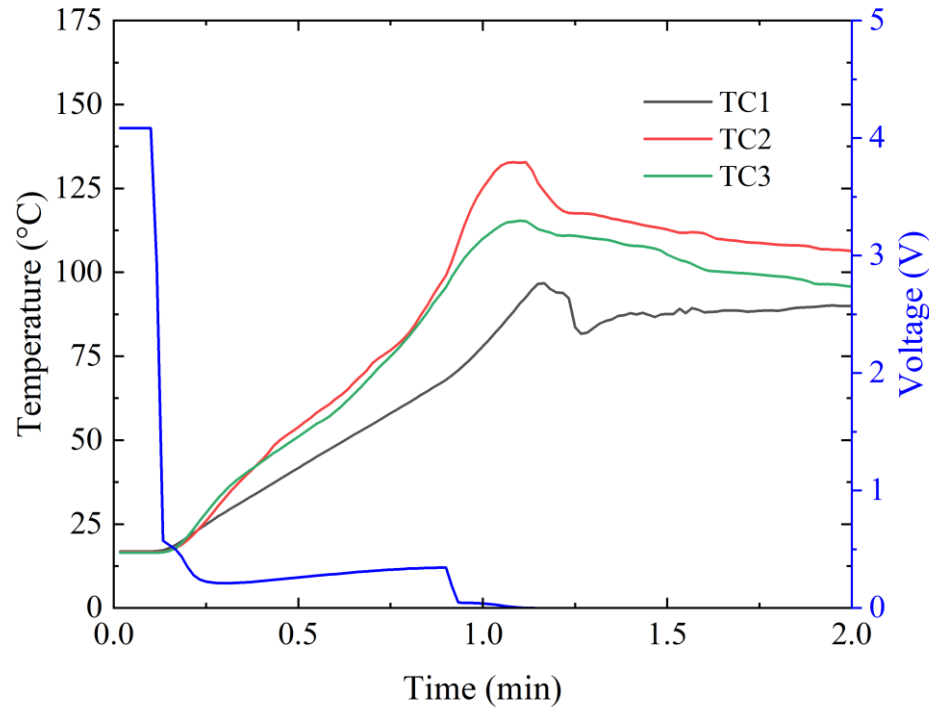


# External Short Circuit

## 100% SOC, 10 mΩ resistor

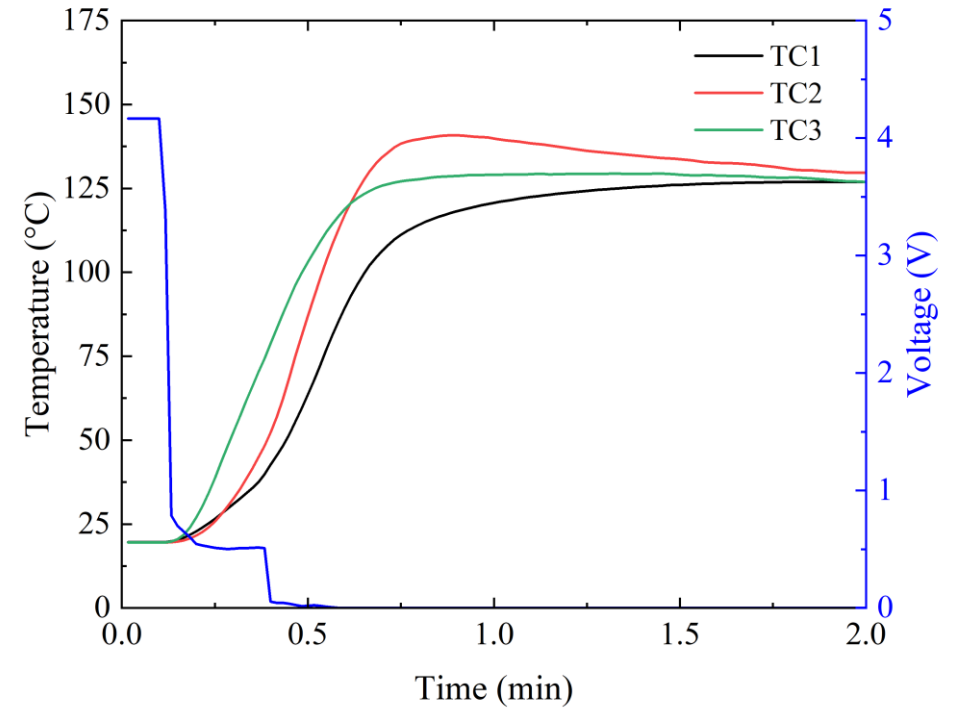


### Na-ion



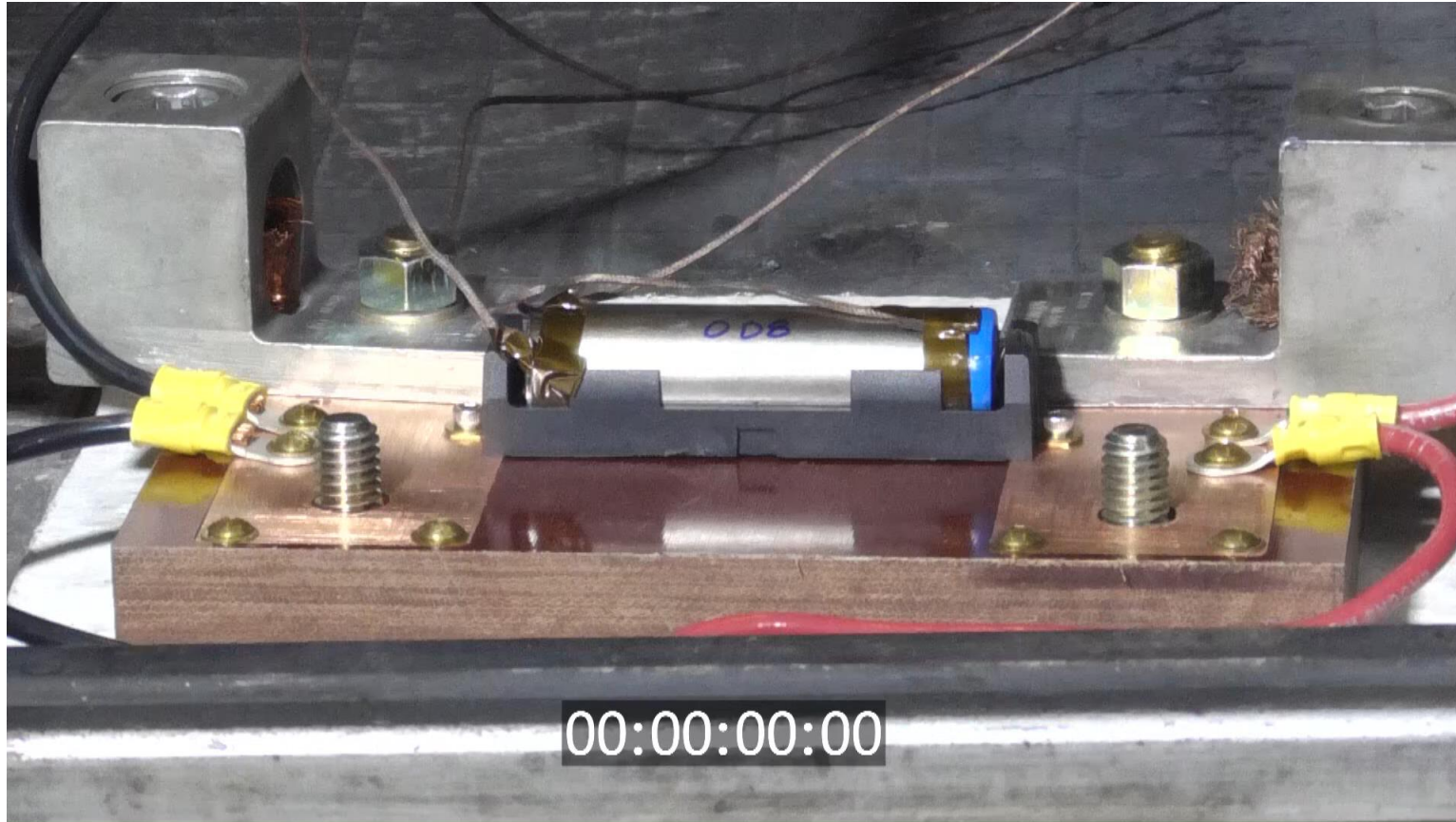
- Max temp: 132.8 °C
- Smoke, vent, electrolyte leak, swelling
- Discharge 159% more capacity than Li-ion

### Li-ion



- Max temp: 140.8 °C
- Swelling

# External Short Circuit, Na-ion

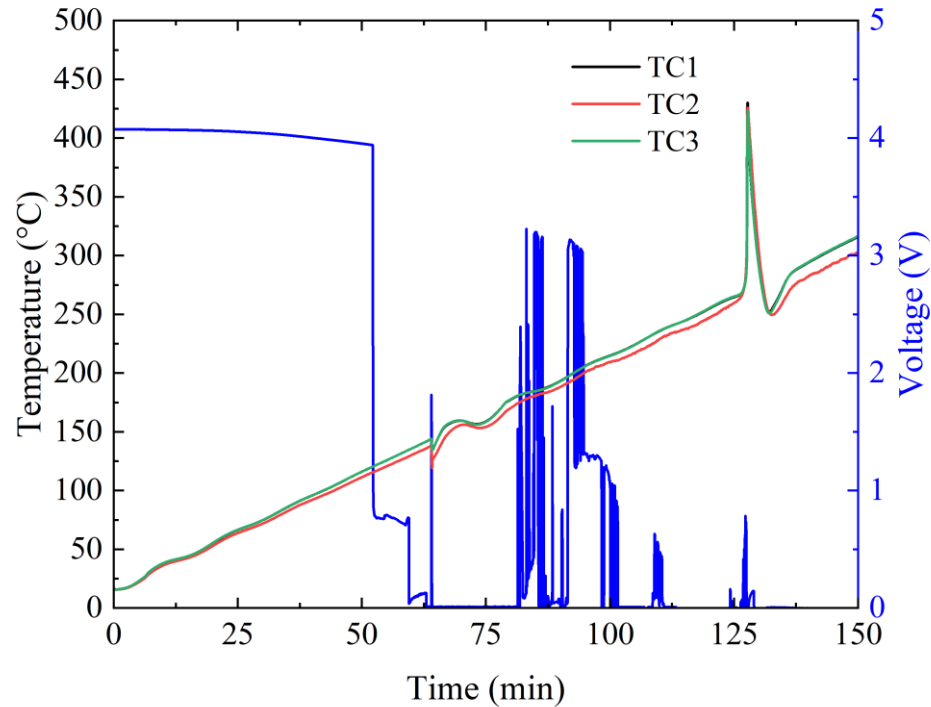


# Thermal Ramp

100% SOC, 2 °C/min, 500 °C or Failure

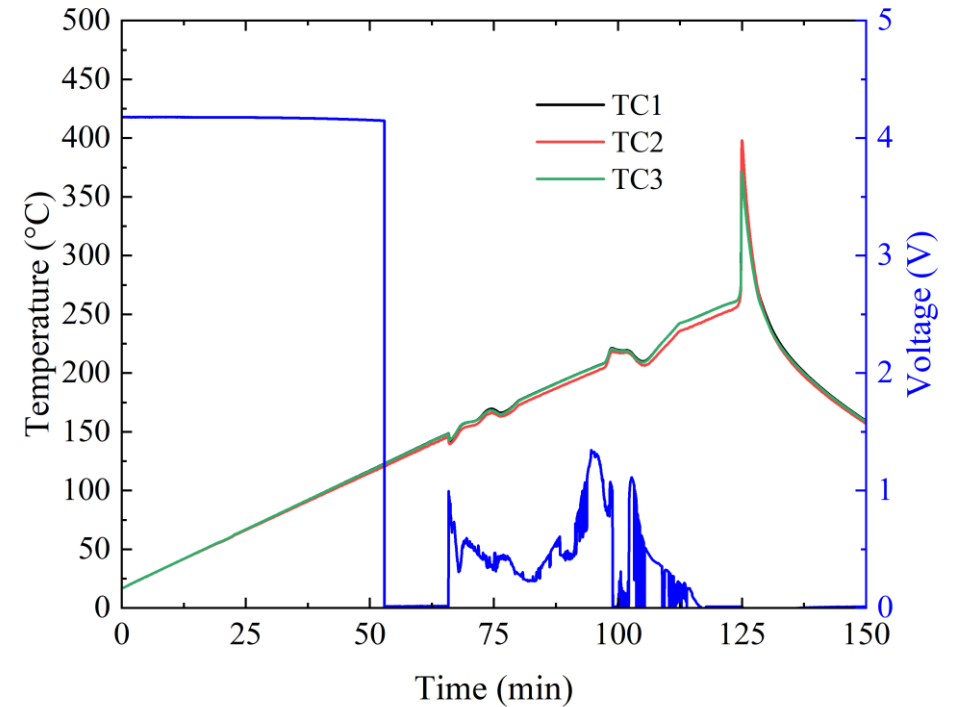


Na-ion



- Max temp: 430.1 °C
- Voltage drop: 120.4 °C
- Vent: 143.9 °C
- Thermal runaway (>10 °C/min): 271.1 °C
- Vent, electrolyte leak

Li-ion



- Max temp: 398.1 °C
- Voltage drop: 122.2
- Vent: 148.4 °C
- Thermal runaway (>10 °C/min): 269.1 °C
- Vent, electrolyte leak

# Thermal Ramp, Na-ion

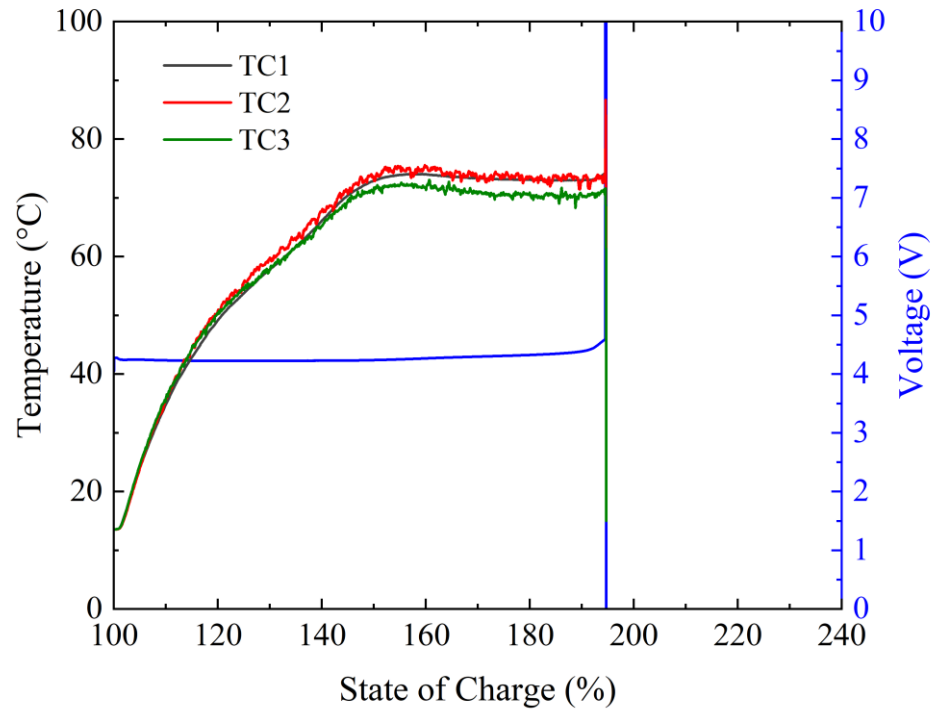


# Overcharge

1C to 250% SOC, Hold for 15 minutes or Failure



Na-ion



Post Overcharge Crush



- Max temp: 75.4 °C
- Compliance voltage: 194.5% SOC
- No observable event

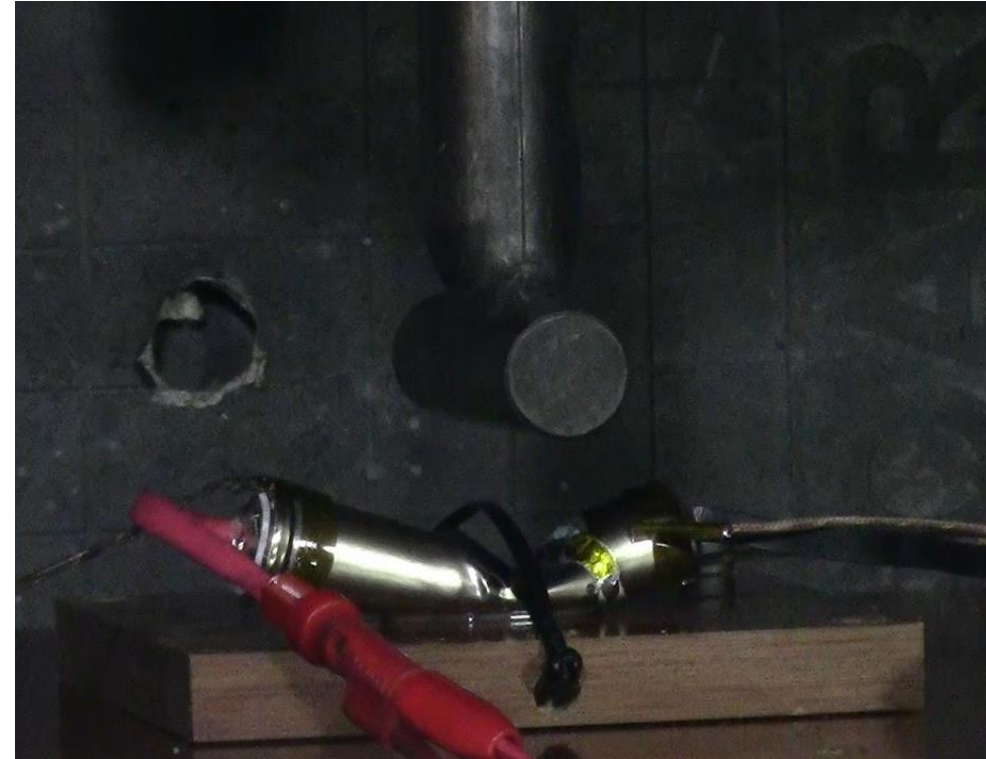
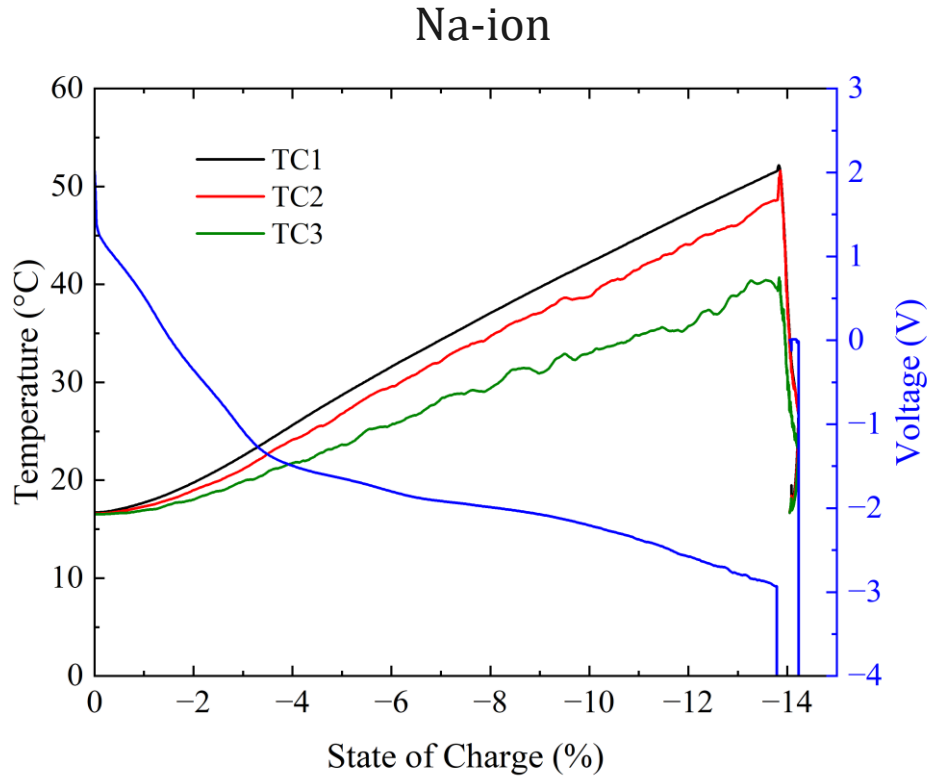
Cell had no voltage but, energy remained.

# Overdischarge

1C to -150% SOC, Hold for 15 minutes or Failure



*Post Overdischarge Crush*



- Max temp: 52.1 °C @ -13.8% SOC
- Voltage polarization: -13.8% SOC
- Realistic practical discharge capacity: 0.023 Ah
- No observable event

Liquid electrolyte leakage.  
IR camera showed no signs of heating.

# Comparison Summary



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	Li-ion	98.1	1.88 mm	No observable effect
Crush	Na-ion	167.9	6.67 mm	Vent, smoke, sparks, lid pop
	Li-ion	296.3	9.71 mm	Vent, smoke, sparks, lid pop
External Short Circuit	Na-ion	132.8	N/A	Vent, smoke, electrolyte leak, swelling
	Li-ion	140.8	N/A	Swelling
Thermal Ramp	Na-ion	430.1	271.1 °C	Vent, LE leak
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## ***Key Takeaway***

- ***Na-ion cells have a propensity for destructive failure under mechanical abuse***

# MATERIALS SCALE TESTING

# Case Study– Ford Ecostar

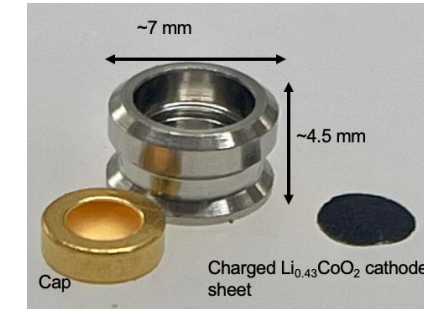
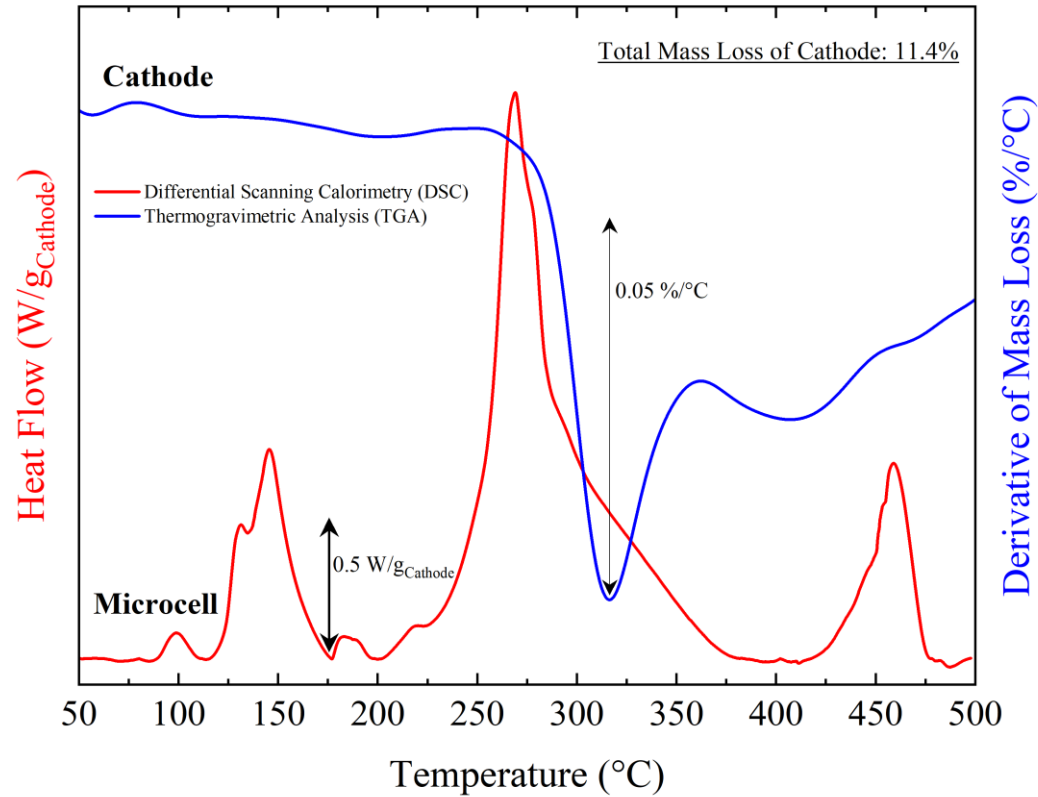
## *Sufficient Safety Research is Necessary Before Commercialization*



- 1965 – Ford developed sodium-sulfur technology
- 1993 – ~100 prototype Ecostars began production
  - Sodium-sulfur battery
    - 332 V architecture
    - 600 °F internal temperature
  - \$250,000 /car (hand built, non-production battery)
- 1993 to 1996 – Two vehicles caught fire while charging
- 1997 – Ford ends sodium-sulfur research

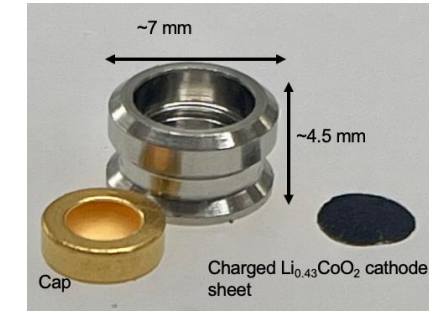
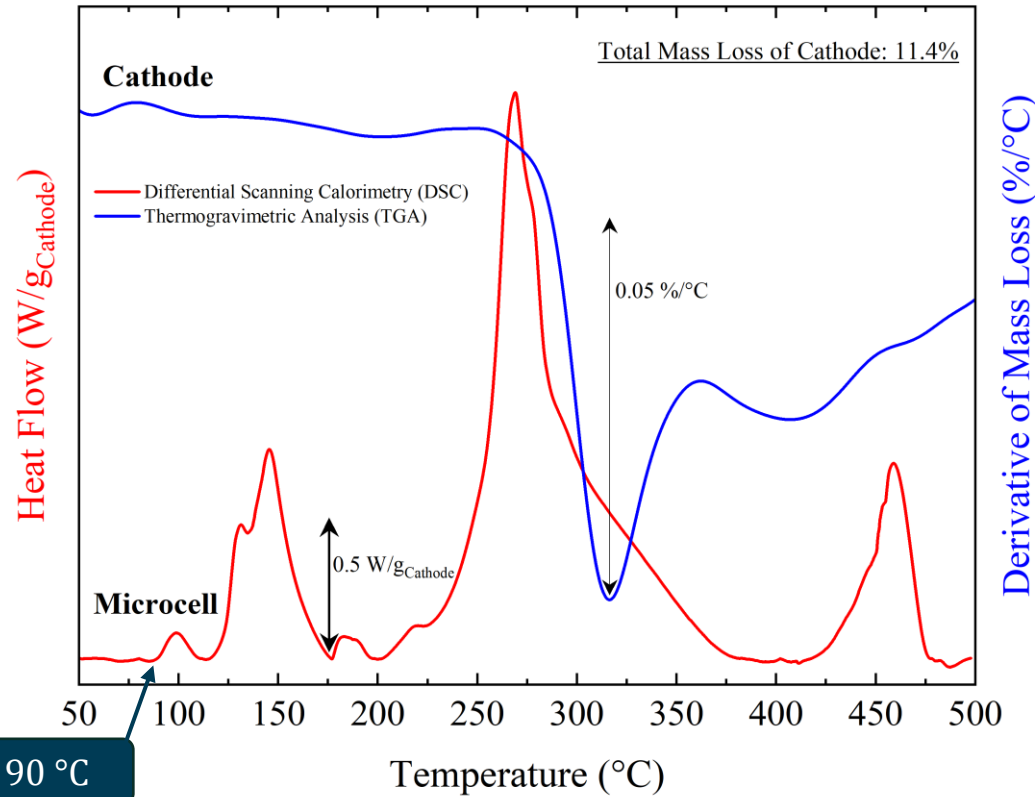


# Methods to Uncover Reaction Pathways



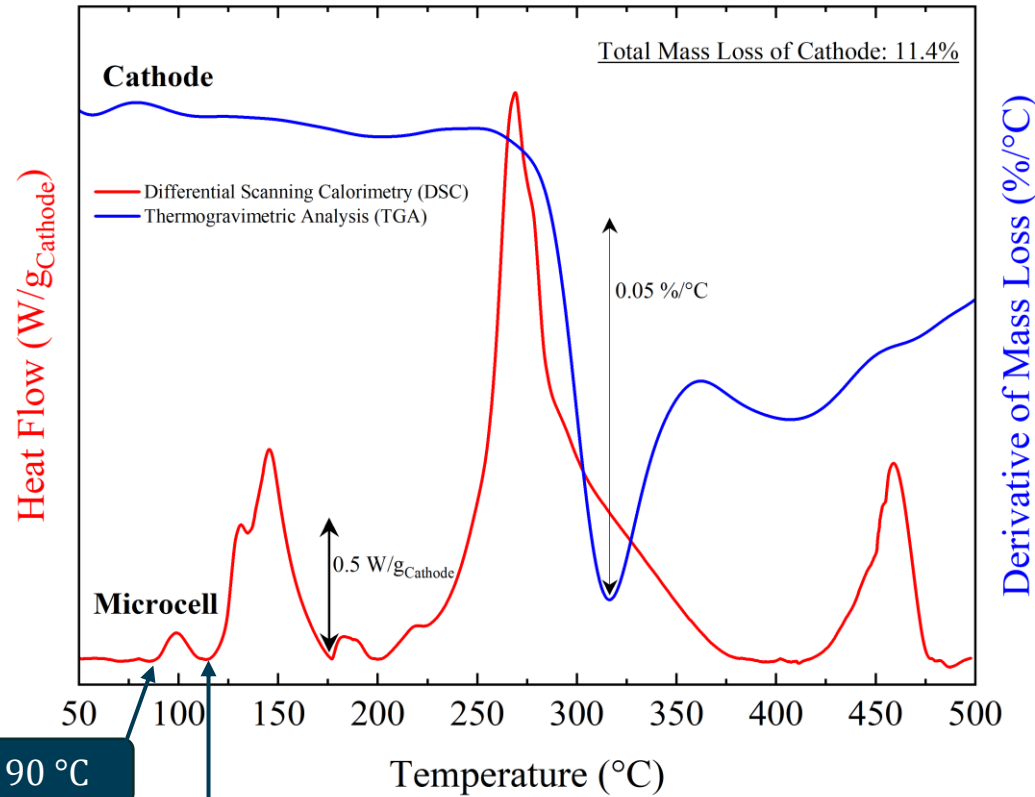
- *Based on these results, we have an understanding of what chemicals are resident to the system.*
- *We can begin to formulate hypothesis on reaction pathways.*

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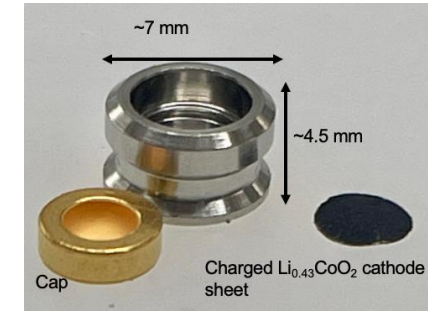
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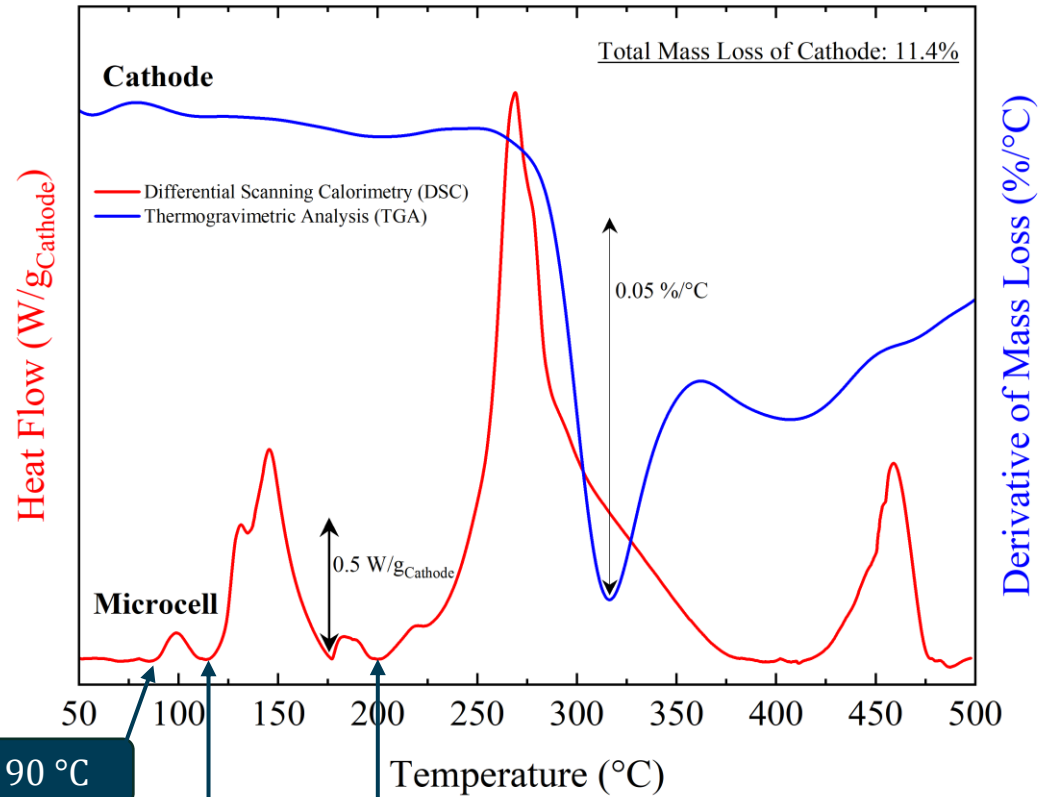
Boiling point DMC: 90 °C

Boiling point EMC: 107 °C



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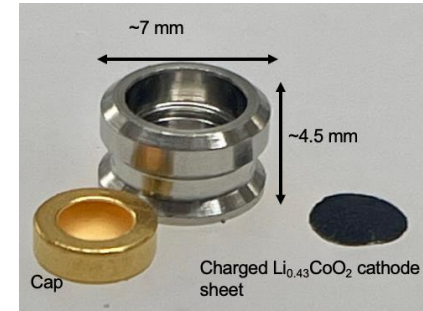
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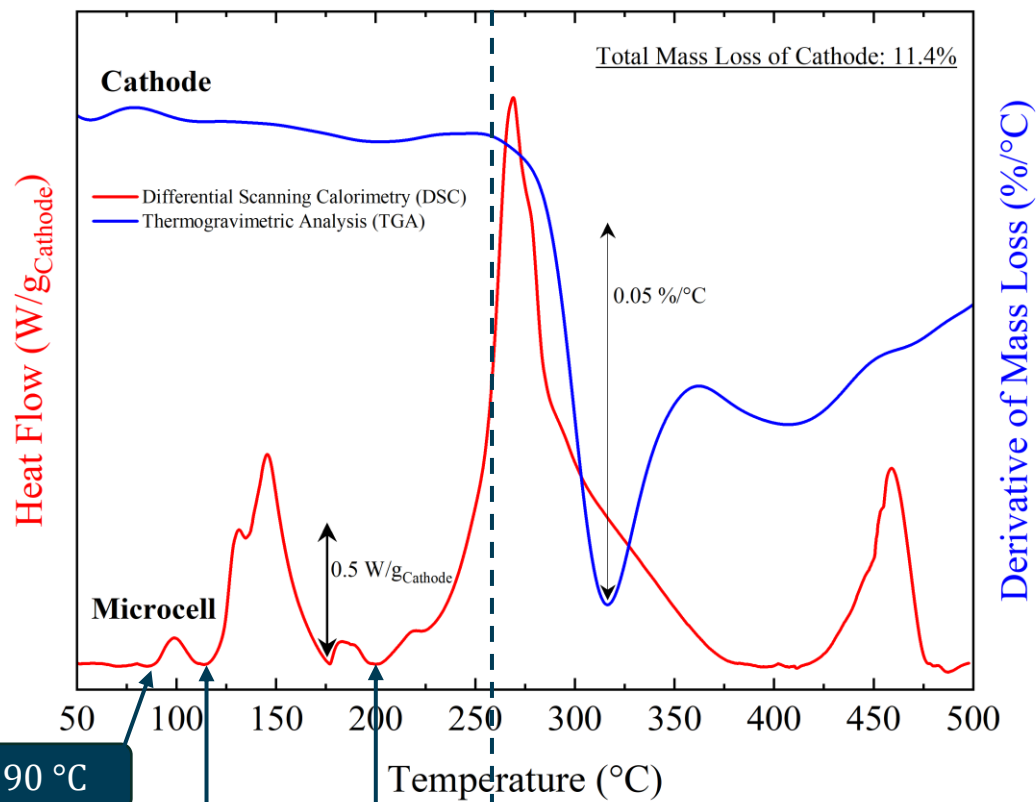
Boiling point EMC: 107 °C

Boiling point PC/EC: 242/248 °C  
Cathode mass loss: ~260 °C



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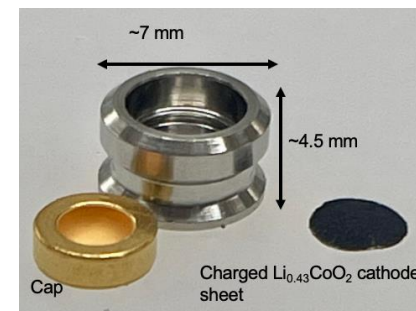
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Boiling point EMC: 107 °C

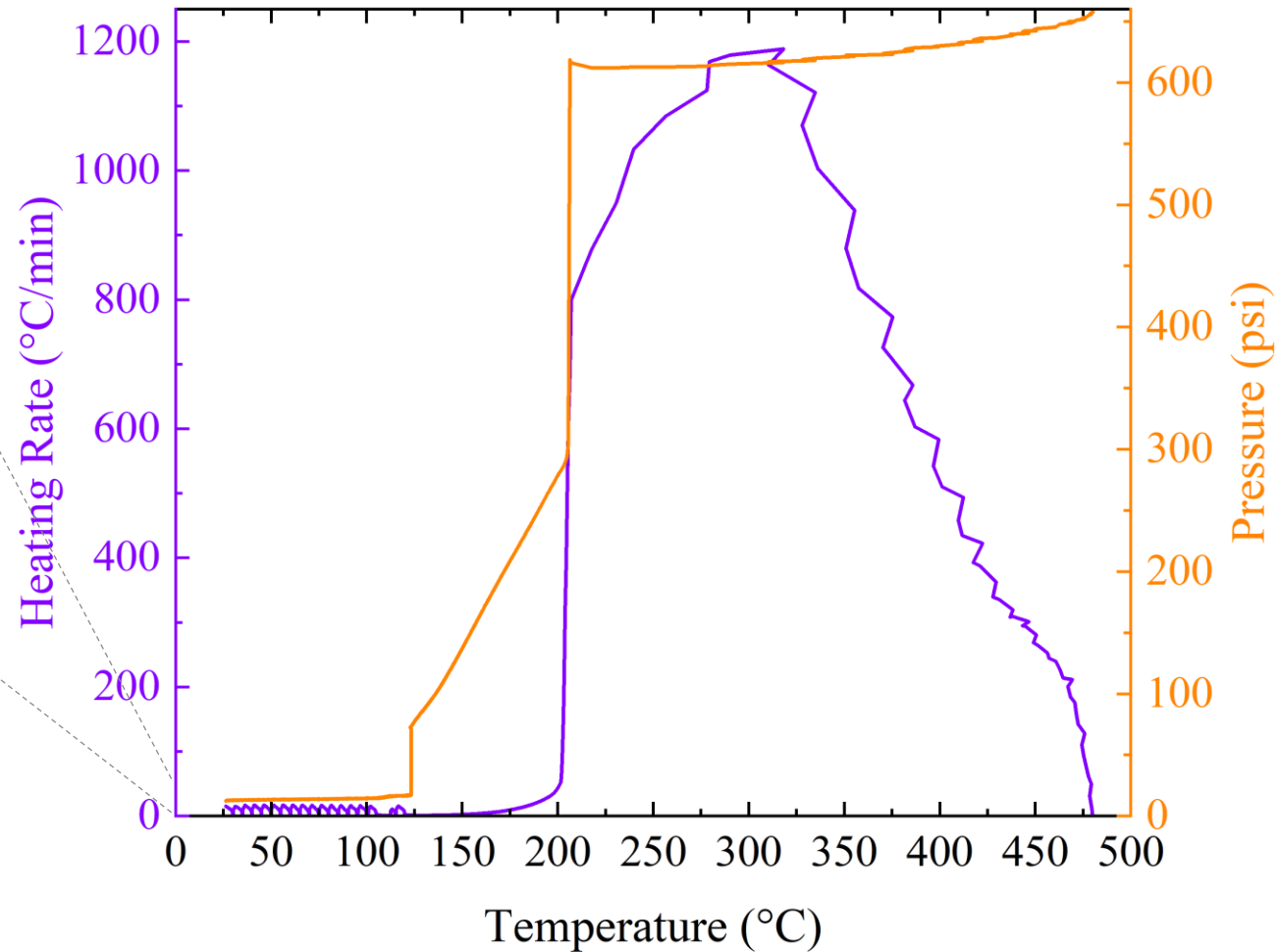
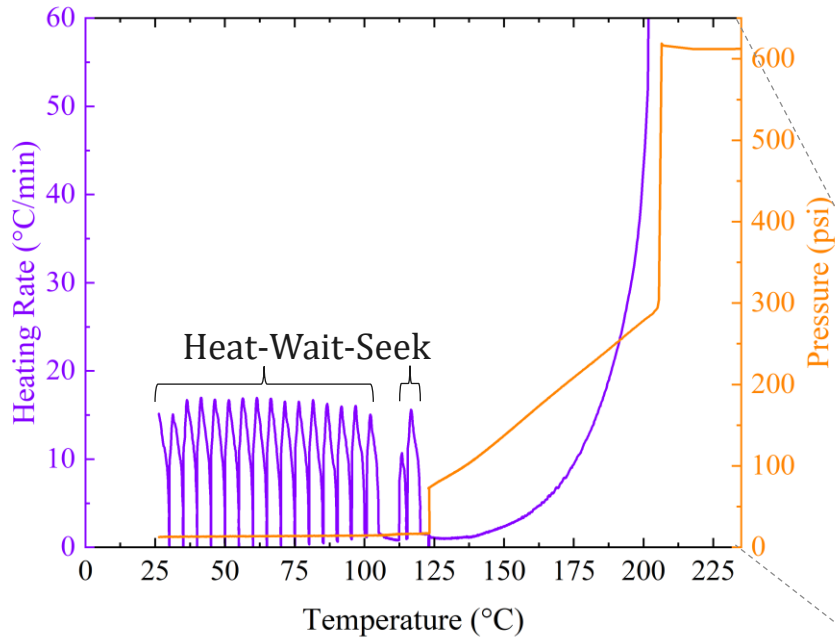
Boiling point PC/EC: 242/248 °C  
Cathode mass loss: ~260 °C



- Based on these results, we have an understanding of what chemicals are resident to the system.
- We can begin to formulate hypothesis on reaction pathways.

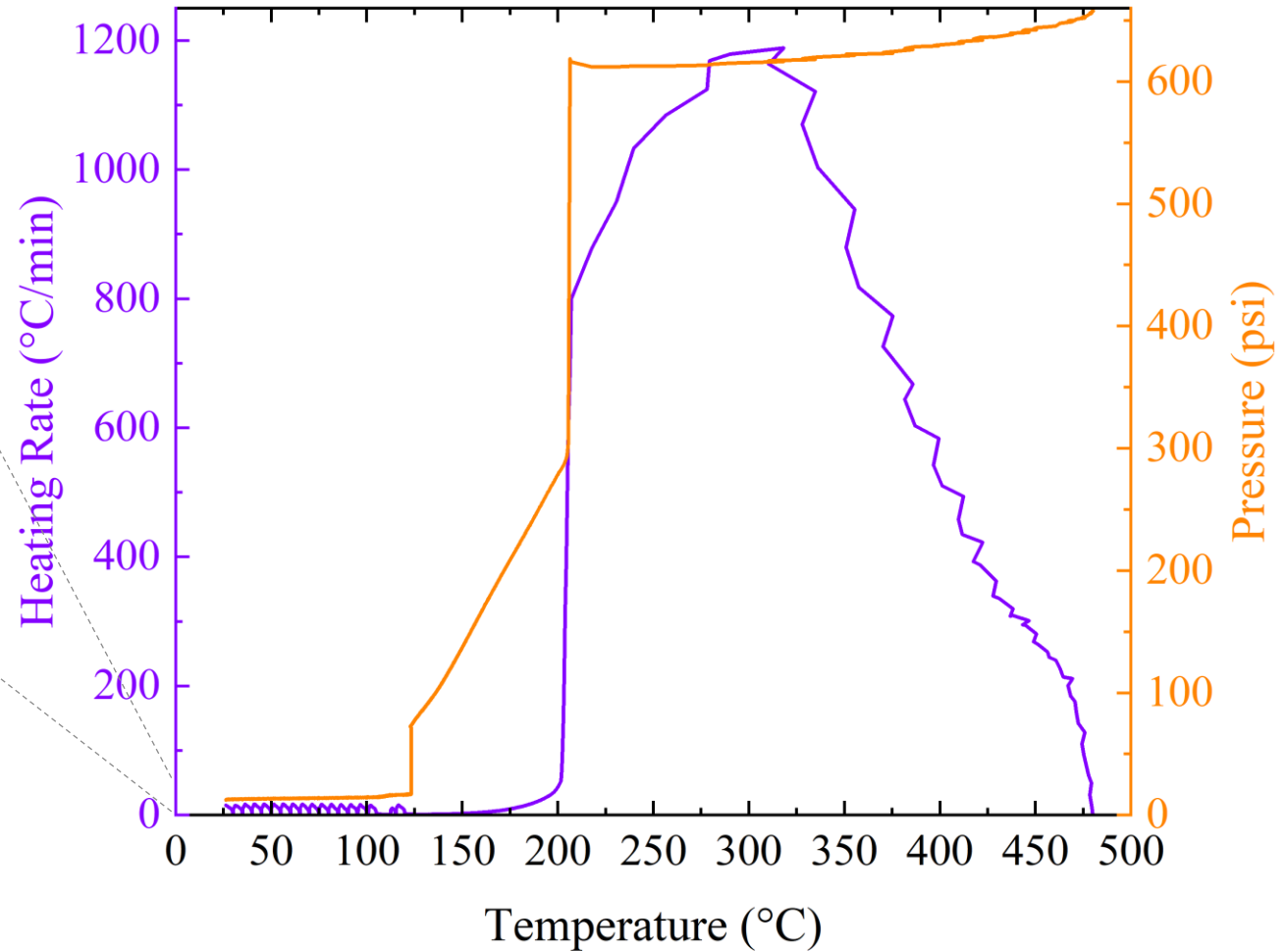
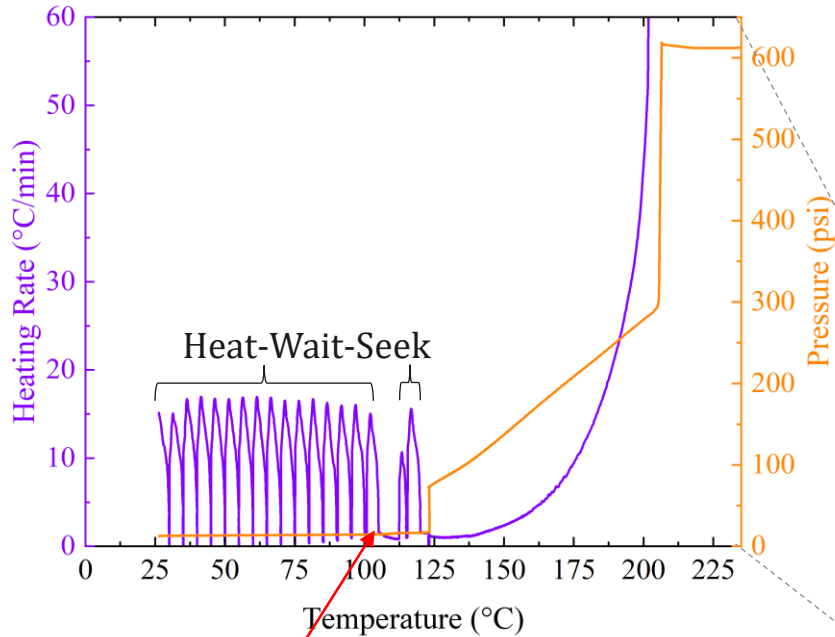


# Accelerated Rate Calorimetry Na-ion Cell



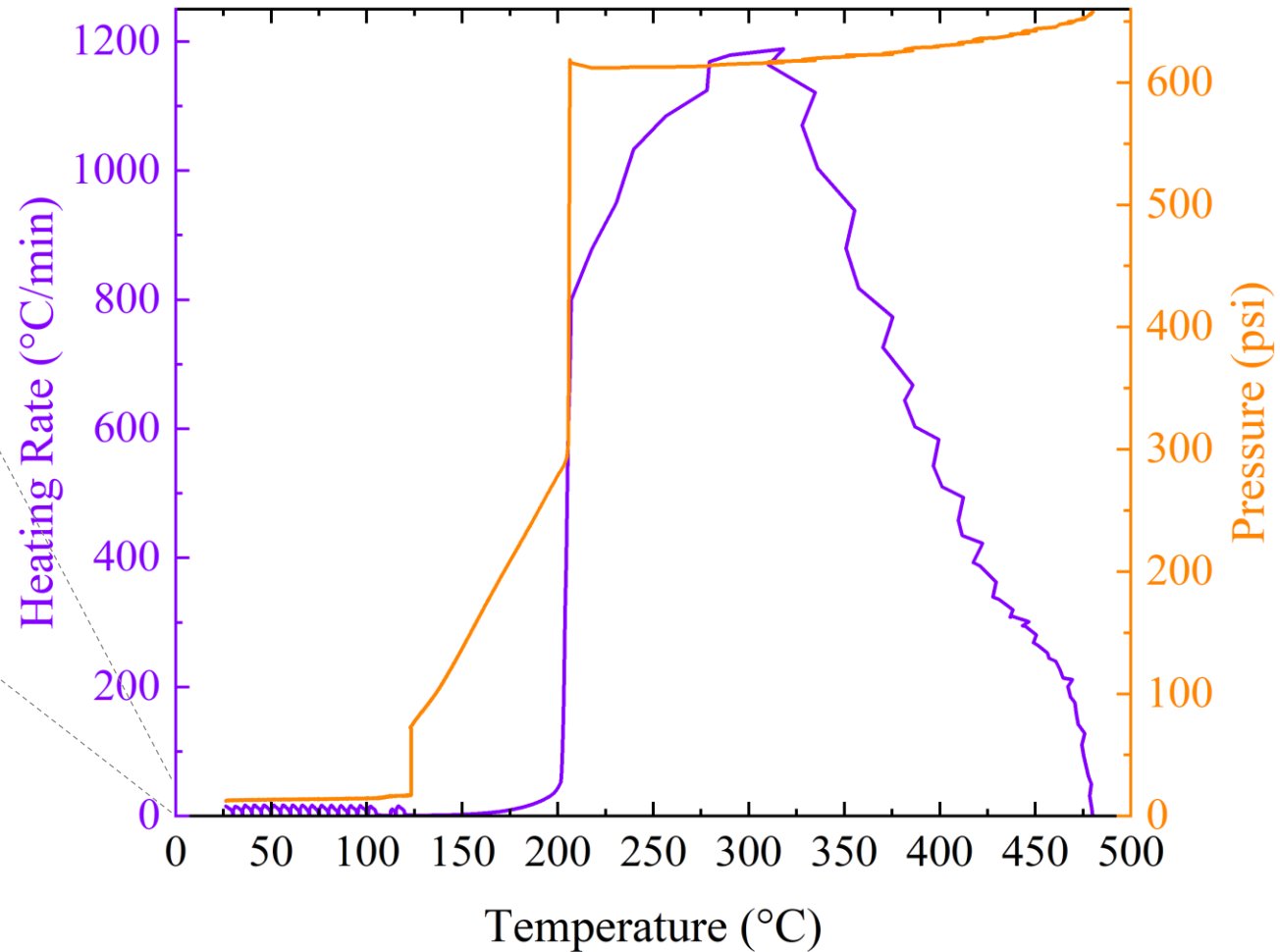
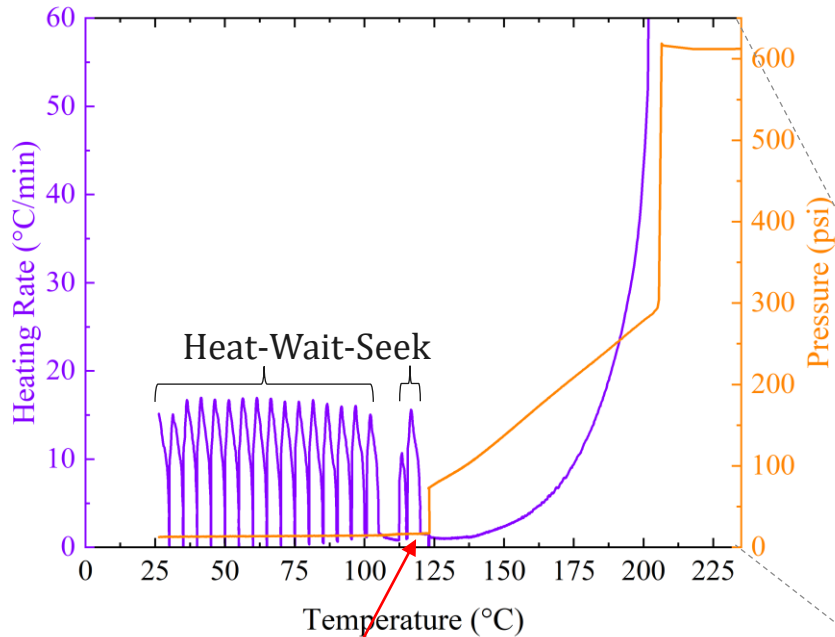
Condition	Temperature (°C)
First self-heating	105.1
Second self-heating	120.3
Vent	123.2
Thermal Runaway (>20 °C/min)	188.6
Pressure Ramp (>300 psi)	205.5
Maximum	480.1

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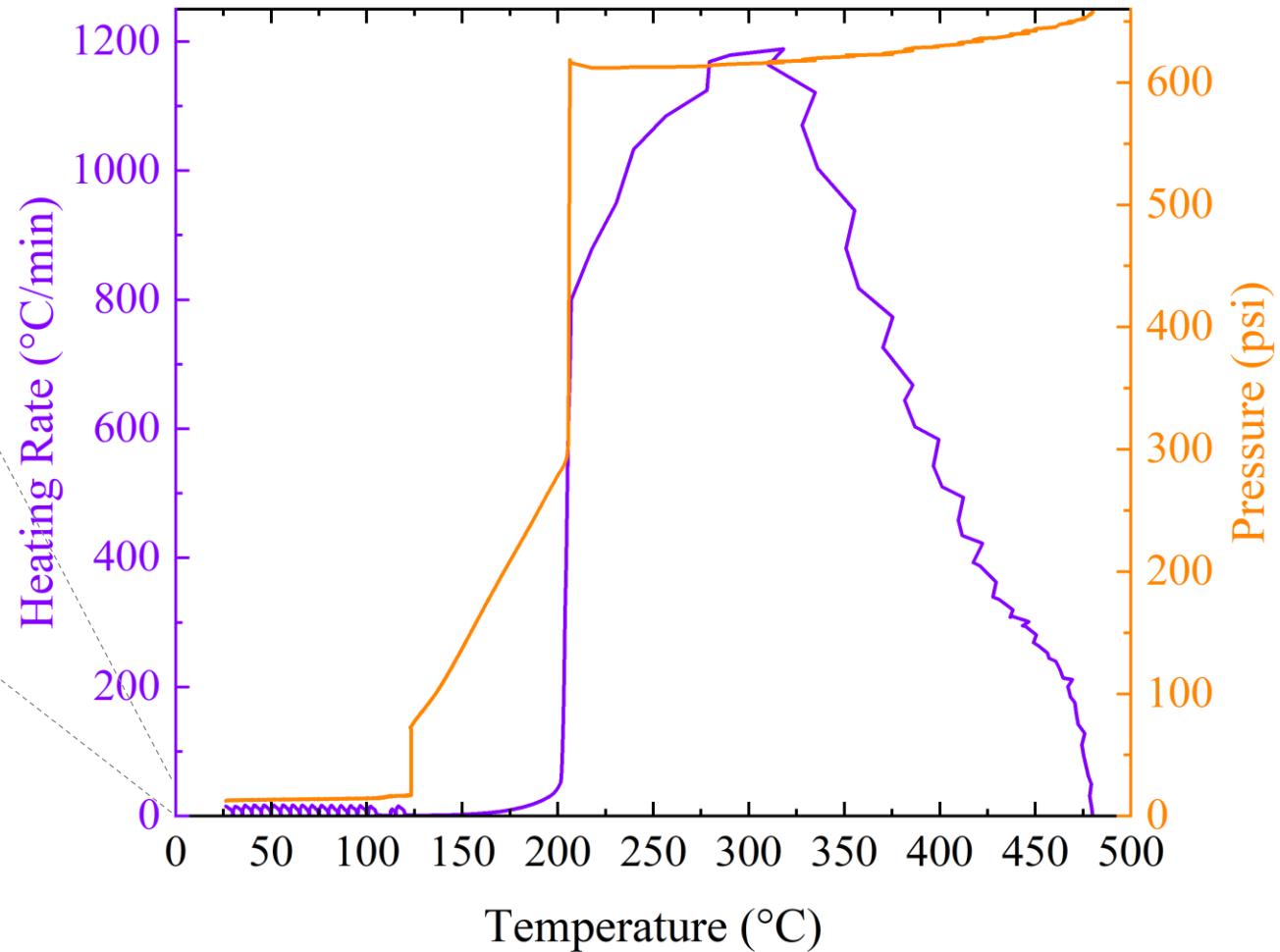
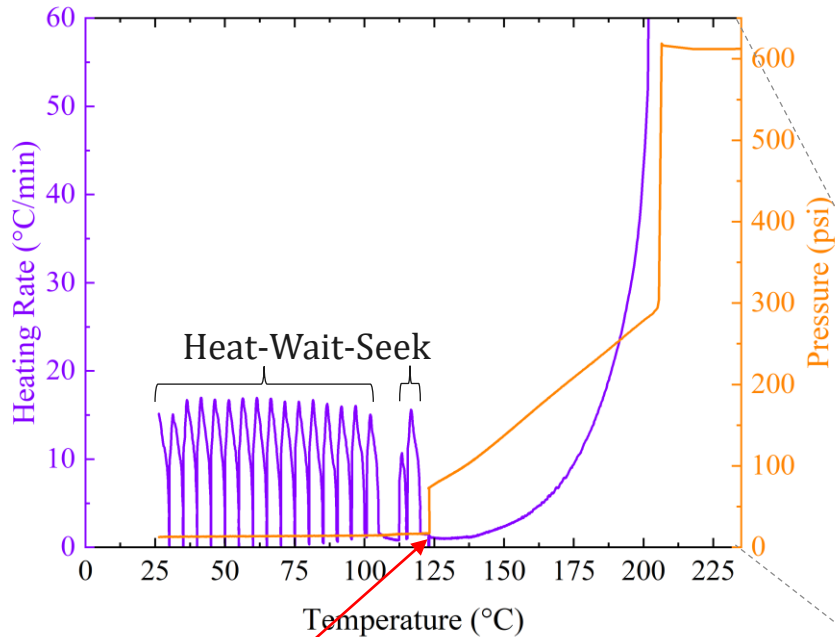
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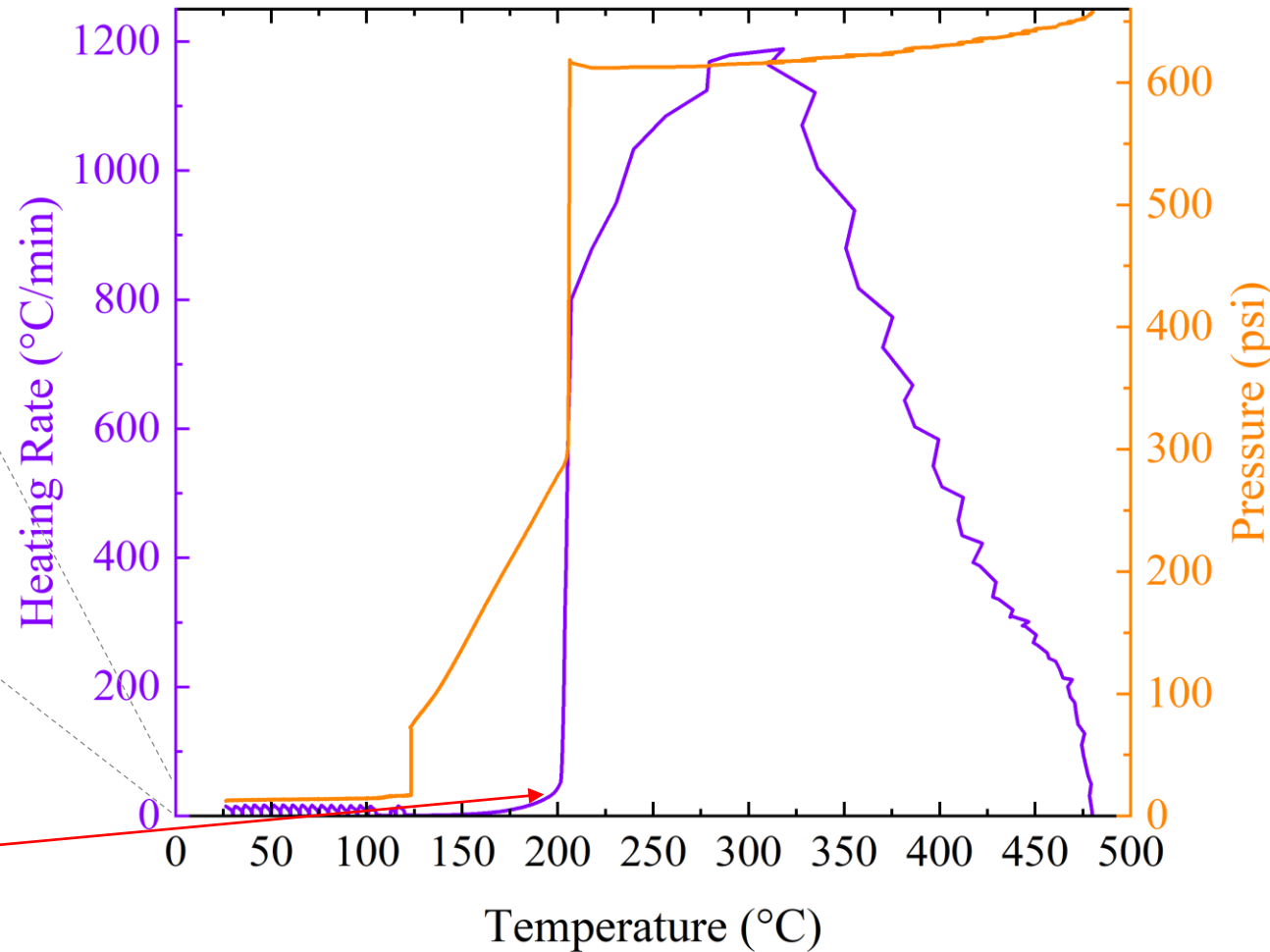
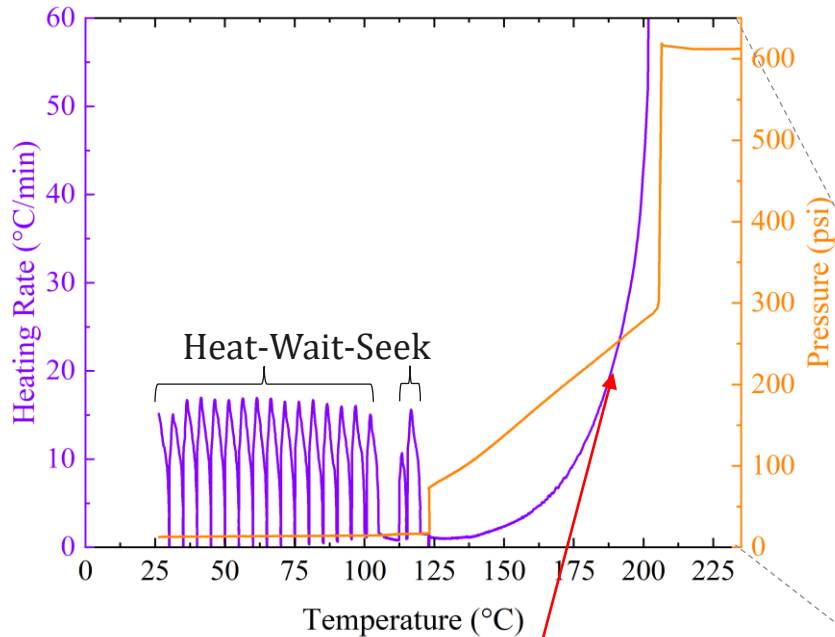
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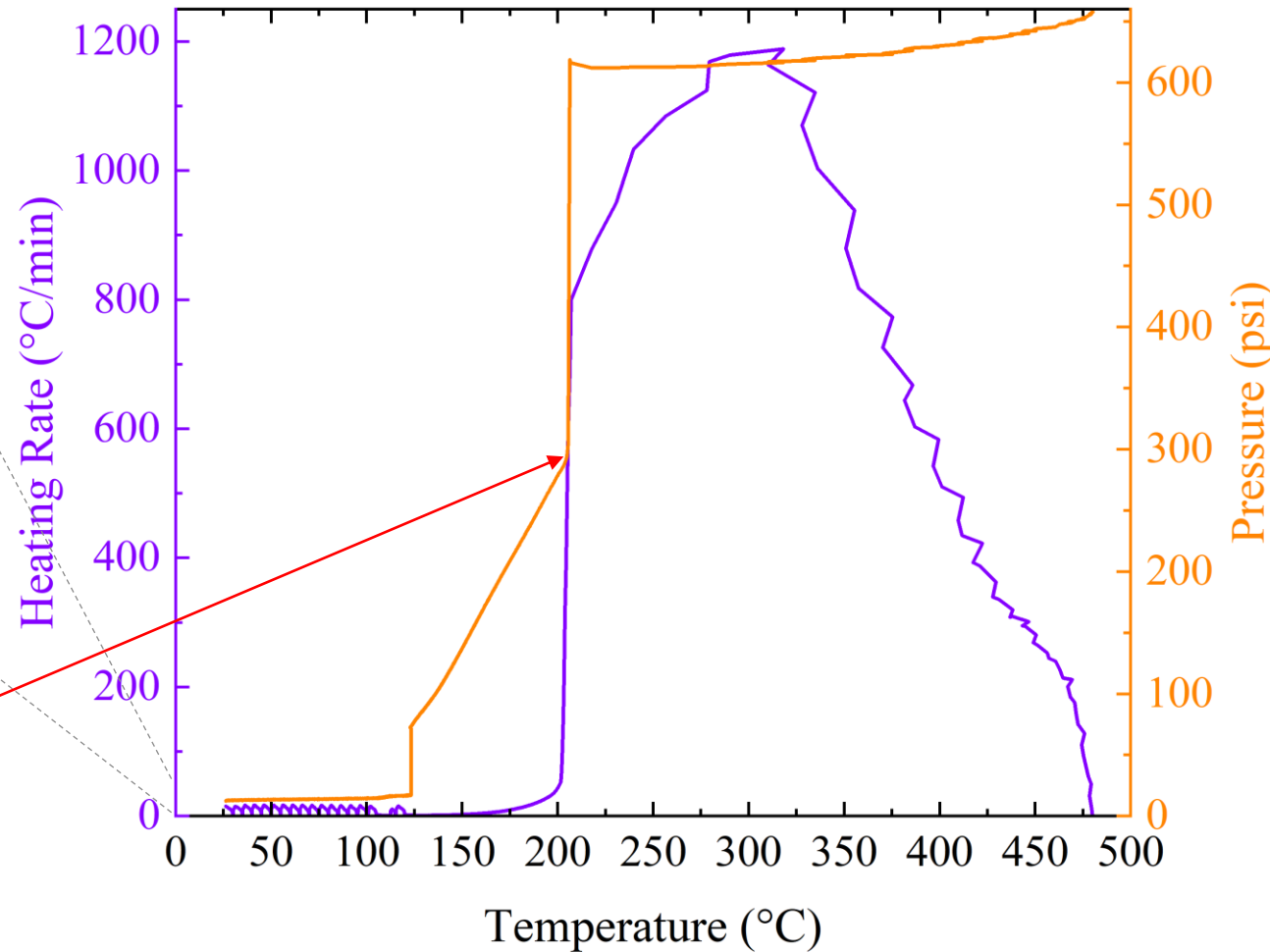
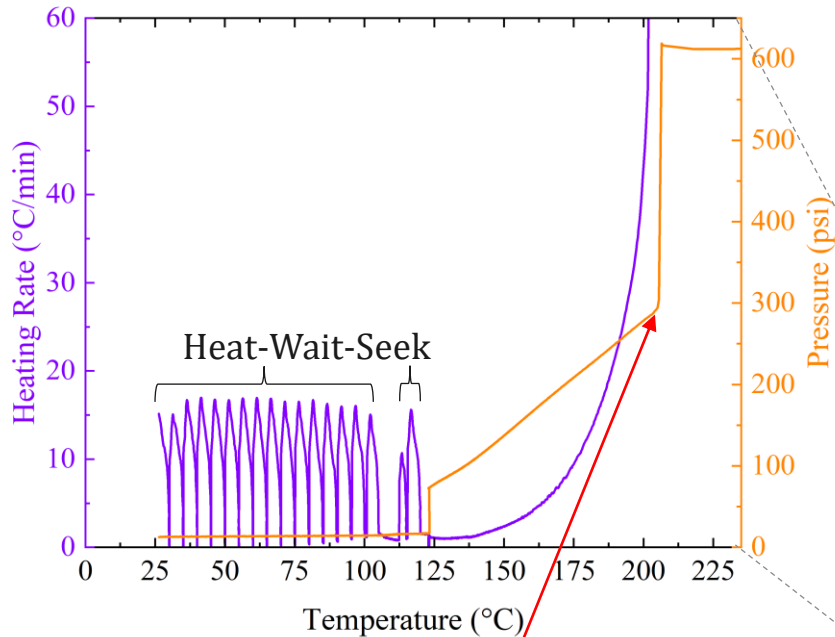
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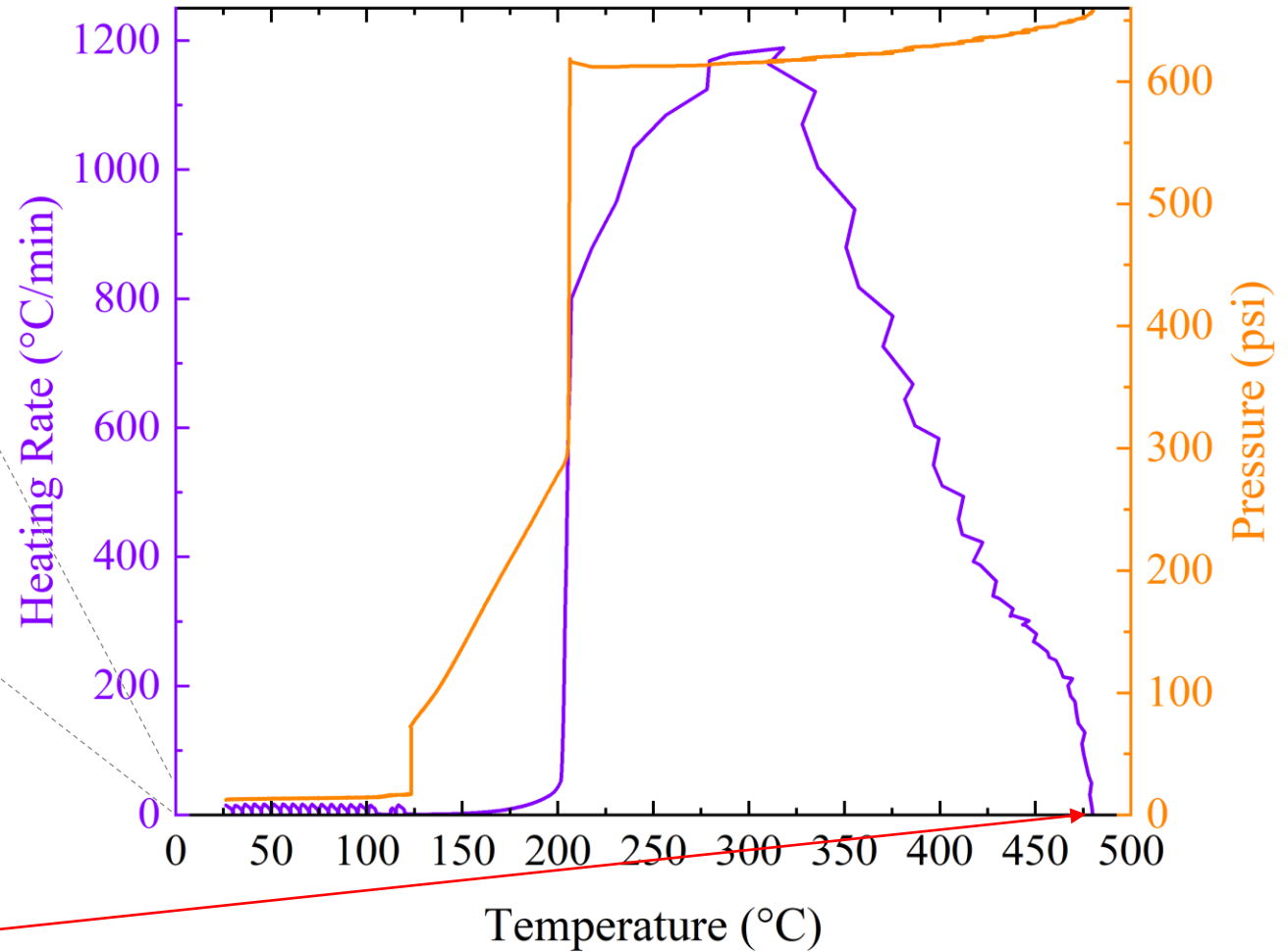
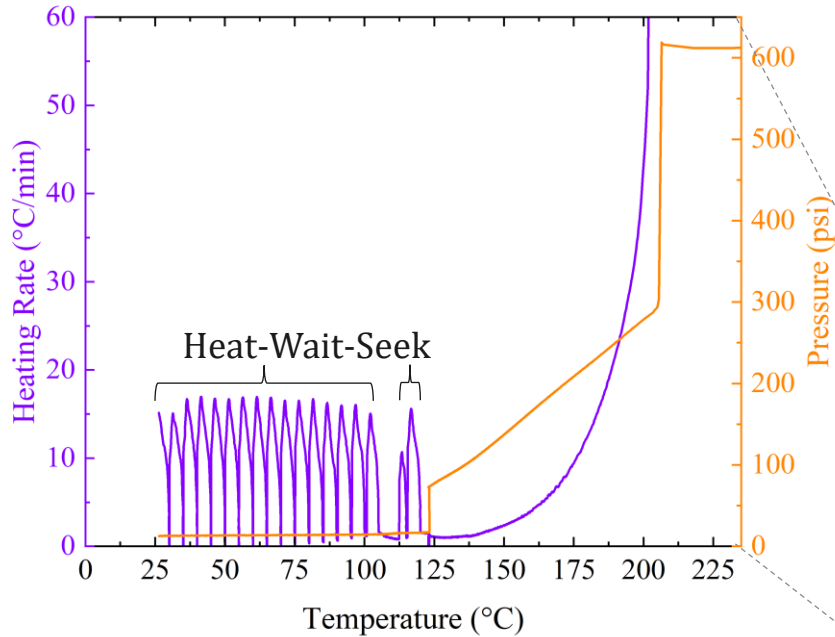
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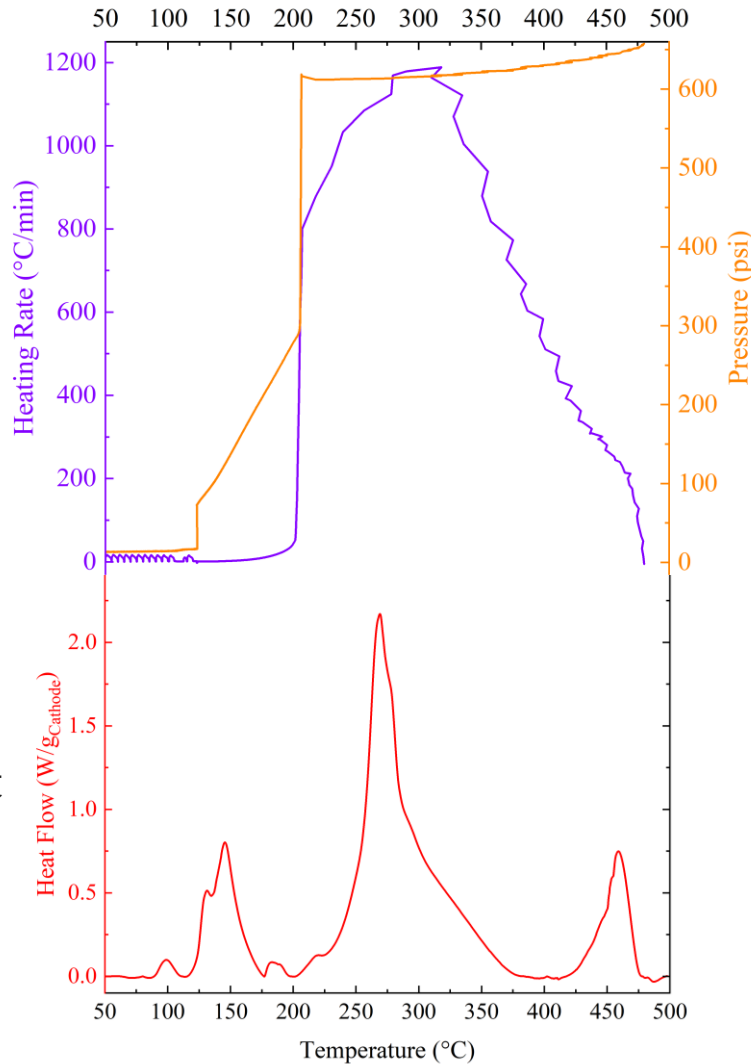
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# Comparison of DSC and ARC Data



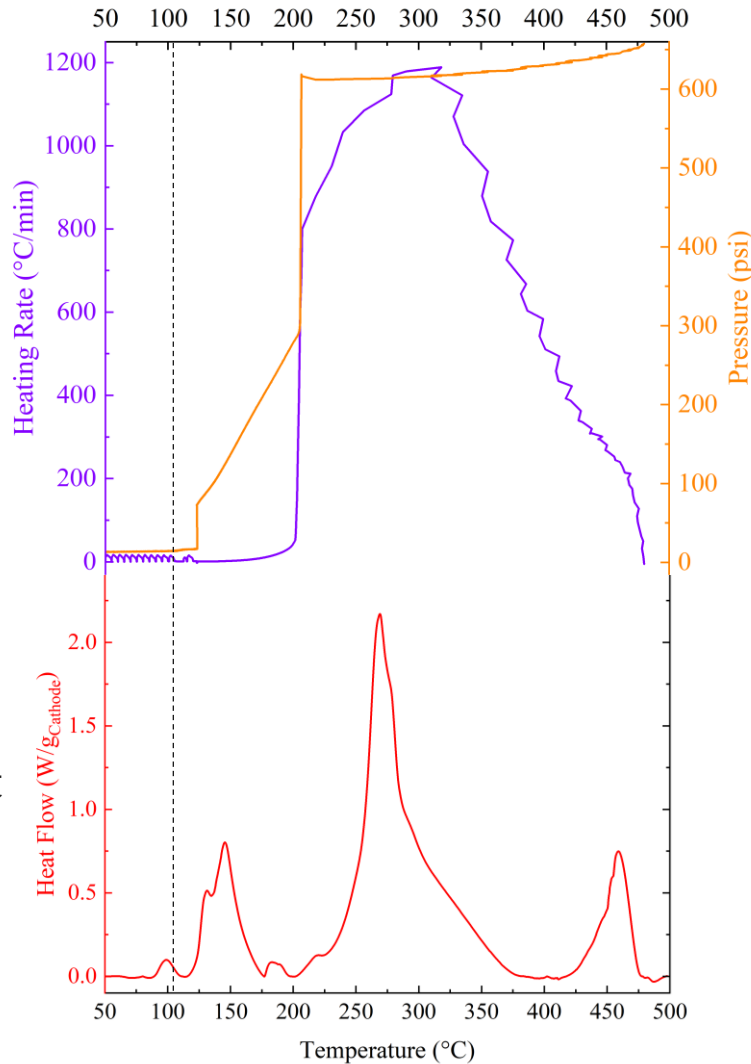
18650 ARC Test

Microcell DSC Test  
(4 mm diameter)

- Initial self-heating corresponds with small exothermic peak
- Return to heat-wait-seek mode in ARC corresponds with return to baseline in DSC
- Onset to high rate heating in ARC corresponds to onset to large exotherm in DSC
- At high temperature, reduction of heating rate in ARC has slowed corresponding with a high temperature exotherm in DSC



# Comparison of DSC and ARC Data

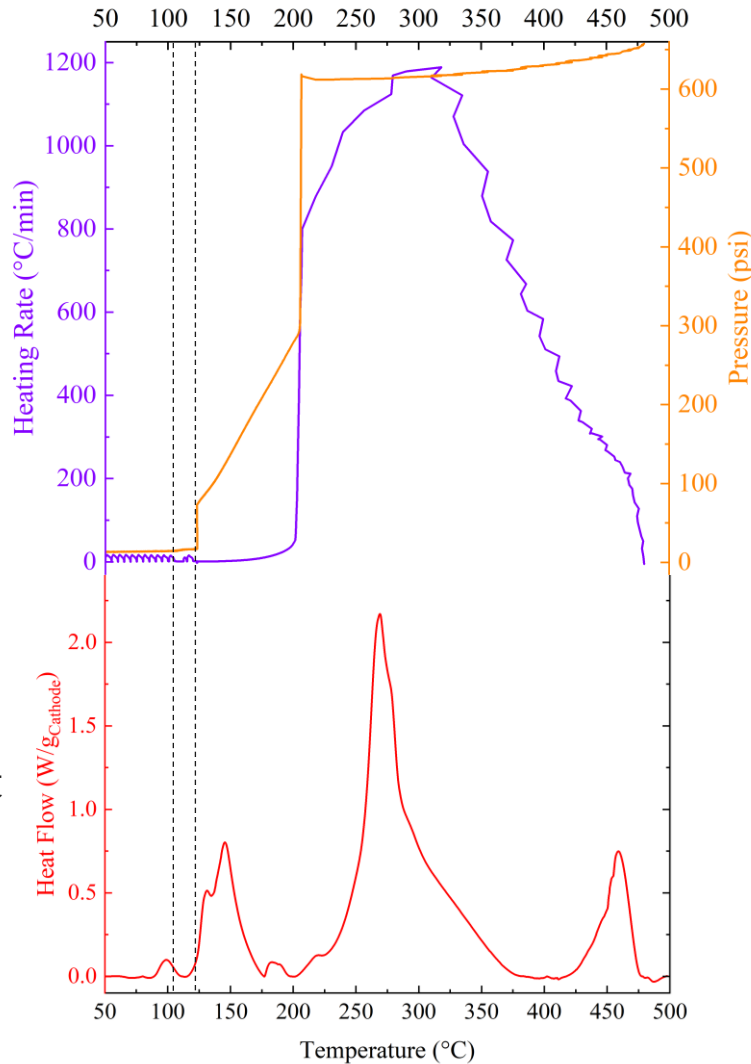


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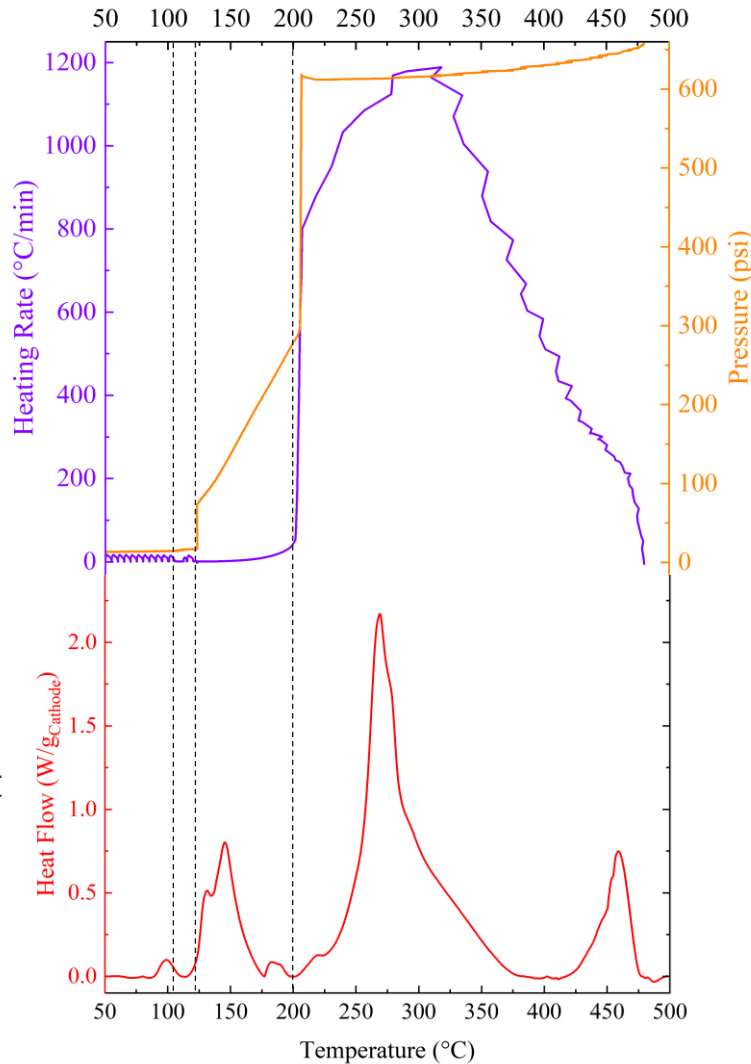


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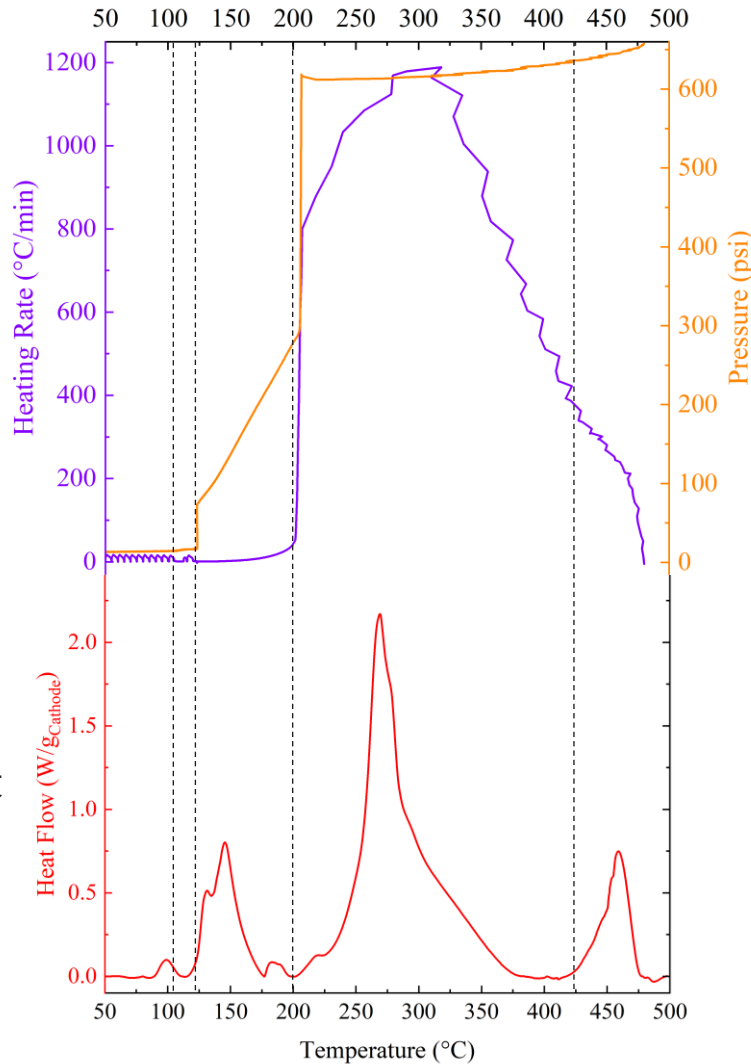


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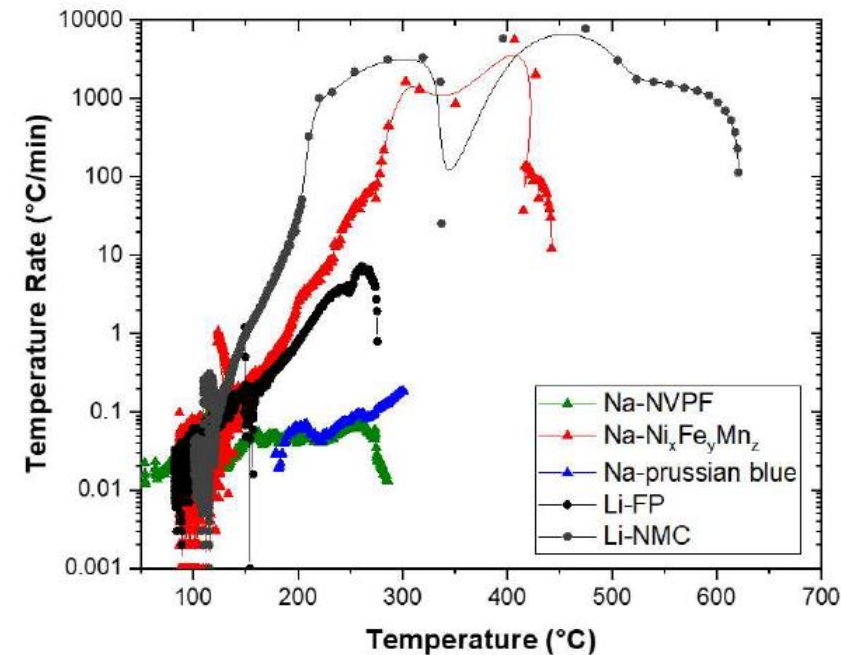
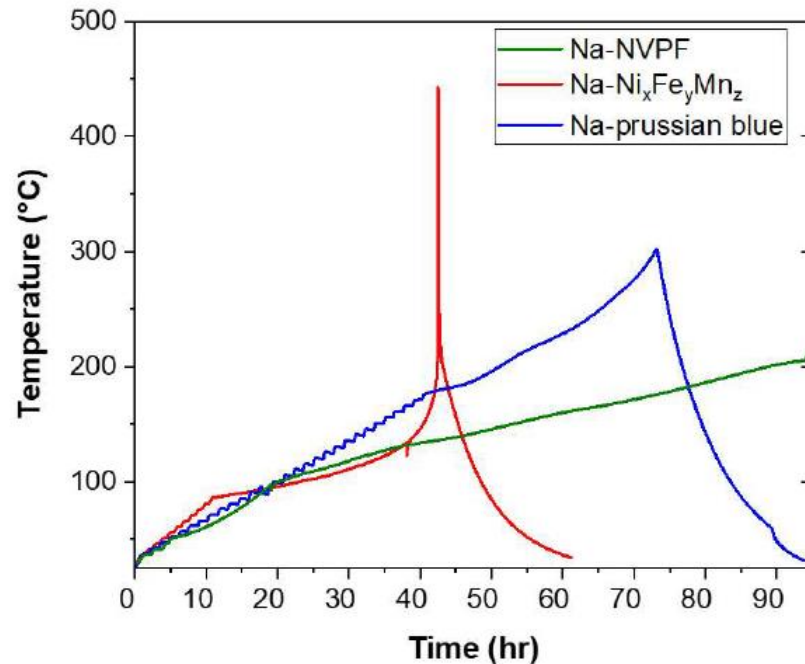
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# Findings At Other Institutions – ARC Results



- US Naval Research Laboratory as presented by Rachel Carter at the 2024 Energy Storage Safety and Reliability Forum, May 16, 2024
  - Red curves come from same manufacturer as SNL cells



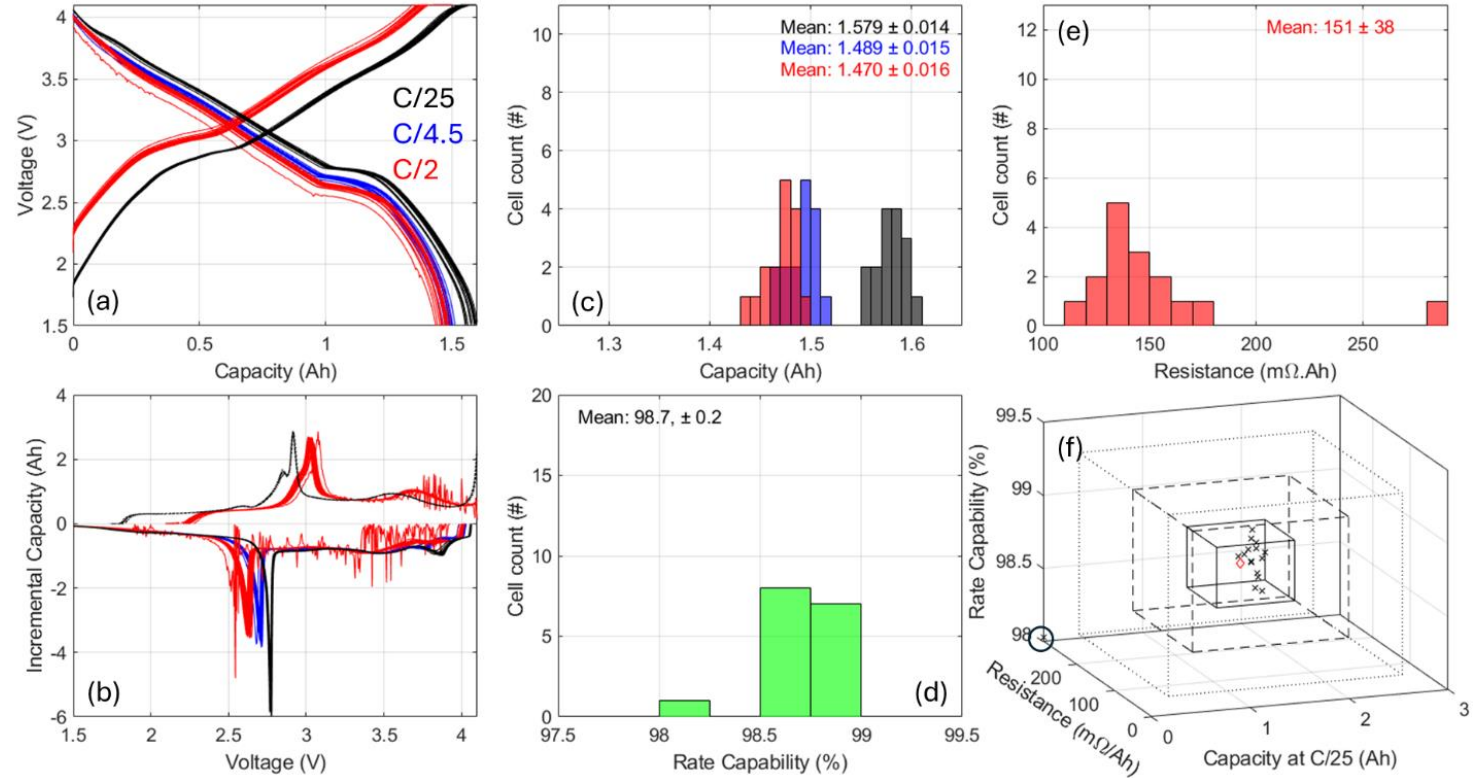
Na-NixFeyMnz exhibits similar thermal runaway to a much higher ED Li-NMC cell but the other Na cells exhibit less thermal event than Li-FP cells

# Cycling Reliability and Degradation



- Reed Wittman, Alex Bates, Loraine Torres-Castro
  - Sandia National Laboratories
- Matthieu Dubarry
  - University of Hawai'i at Manoa
- David Anseán
  - University of Oviedo, Spain

## Cell-to-Cell Variation



(a) Voltage vs. capacity curves at C/25, C/4.5, and C/2 for all the tested HKD cells with. (b) the associated incremental capacity curves, (c) their capacity distribution, (d and e) the rate capability and resistance distributions respectively, and (f) the summary of the cell to cell variations. The inner square represents a  $1\sigma$  spread, the dashed square  $2\sigma$ , and the dotted one  $3\sigma$ .

# SNL Na-ion Safety Team



Alex M. Bates



Loraine Torres-Castro



Nathan Johnson

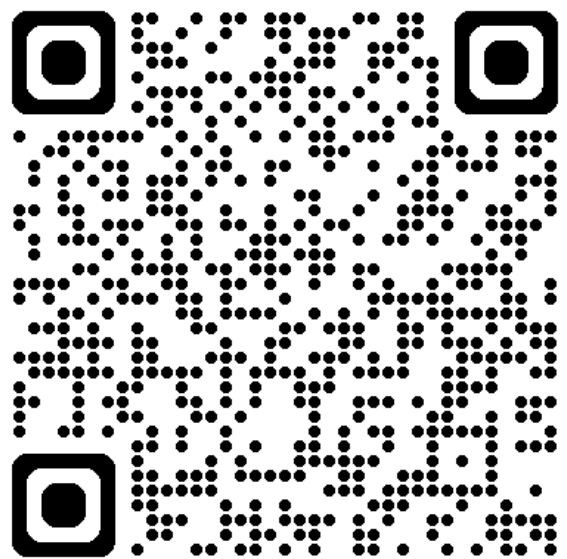
# Acknowledgements



This material is based upon work supported by the U.S. Department of Energy, Office of Electricity (OE), Energy Storage Division.

Sandia National Laboratories is a multi-mission laboratory managed and operated by National Technology and Engineering Solutions of Sandia, LLC., a wholly owned subsidiary of Honeywell International, Inc., for the U.S. Department of Energy's National Nuclear Security Administration under contract DE-NA-0003525. This paper describes objective technical results and analysis. Any subjective views or opinions that might be expressed in the paper do not necessarily represent the views of the U.S. Department of Energy or the United States Government.



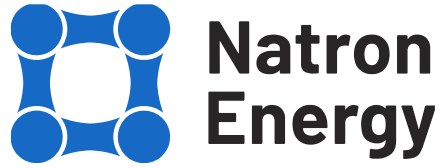


# Questions?

Alex Bates

[ambates@sandia.gov](mailto:ambates@sandia.gov)

<https://www.linkedin.com/in/alex-bates/>

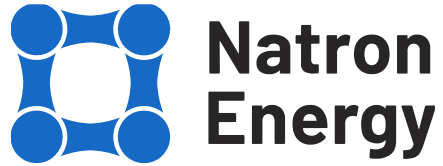


# Introduction

November 2024



*Powered by Prussian Blue*



# Who Is Natron Energy?

## Sodium-ion Batteries Powered by Prussian Blue

### Company:

- Founded in 2012 as a Stanford spin out
- 180+ employees
  - Headquarters - Santa Clara, CA.
  - Manufacturing Plant - Holland, MI
  - Future Gigafactory #1 – North Carolina, 2027

### Products:

- High power, safe, sustainable batteries having no risk of thermal runaway or fire
- Based on new chemistry supported by more than 40 patents





# Key Value Propositions

# The Problems With Many Batteries



Fire



Human and Environmental  
Abuse in Materials Sourcing



Hazardous  
Materials



Ship with minimal charge

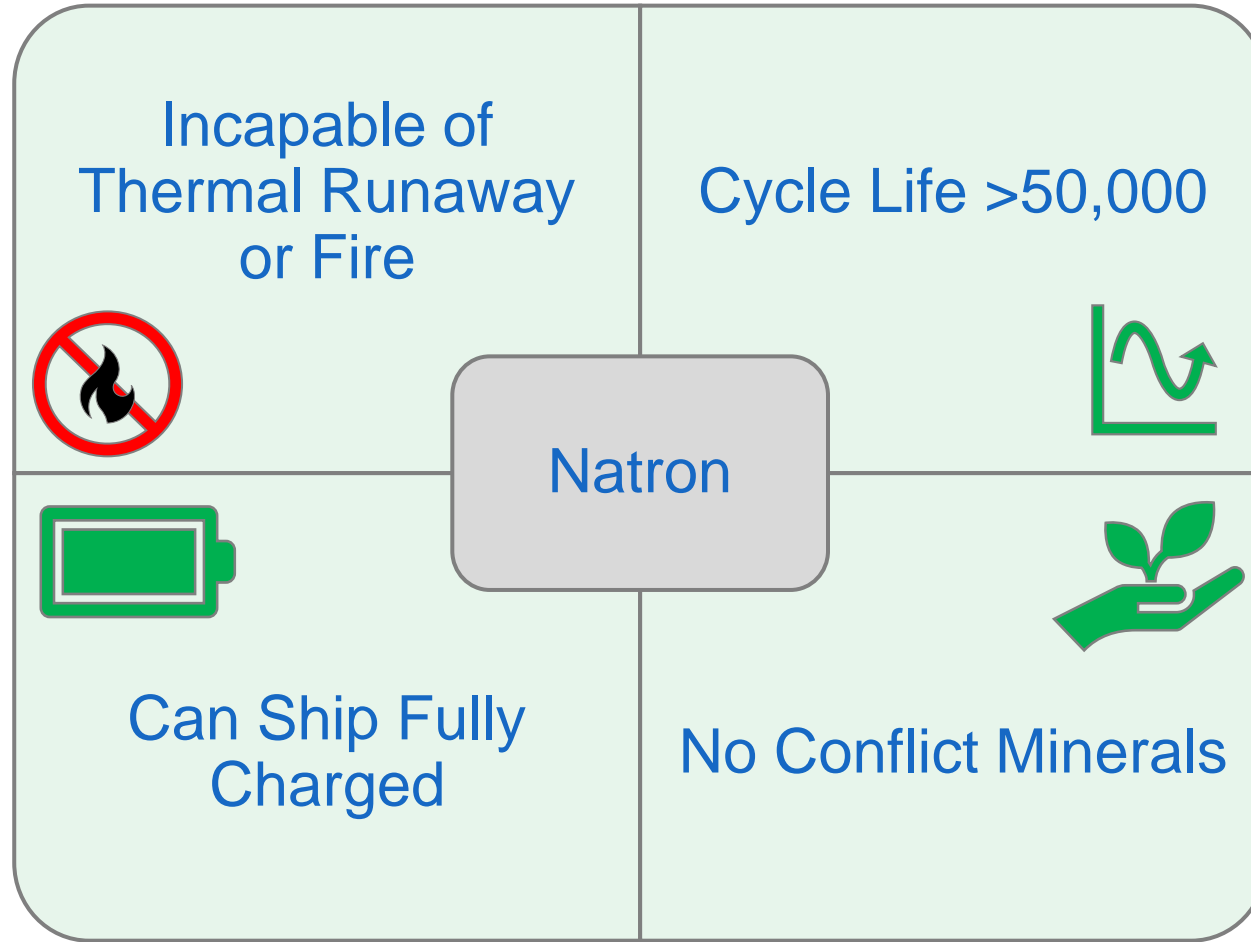


Poison Gas Release



Low Cycle Life

# Prussian blue Sodium-ion breaks new ground in energy storage....



# Sodium-ion Offers Wider Temperature Operating Range

- **Battery Operating Range -20 to 45°C / -4 to 113°F**
  - - 50°C to + 50°C is possible (Consult factory)





# Safety



# Prussian Blue Sodium-ion, The Safest Battery Ever Made!

- No fire or explosion after puncture, pressure, heat, or electrical faults
- Natron is the only battery manufacturer to publish unredacted UL test report

## Nail penetration test

Natron



Li-ion

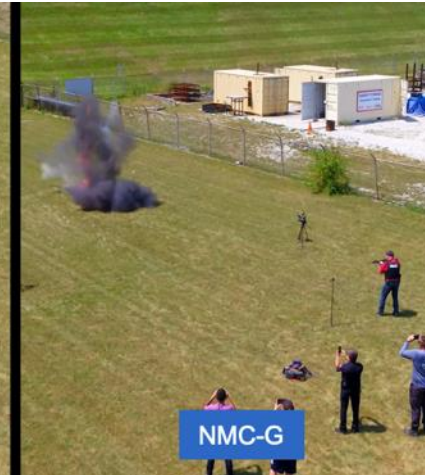


## High speed projectile test

Natron



Li-ion



[Click to view Natron's safety video on YouTube](#)

# Safe To Transport Fully Charged



**Safe for  
Air Transport**

- **Not considered hazardous goods**
- **Can be shipped installed in a battery cabinet**
- **Can be shipped by ground or air fully charged**



*Hazardous warning  
labels not needed  
for Natron batteries!*

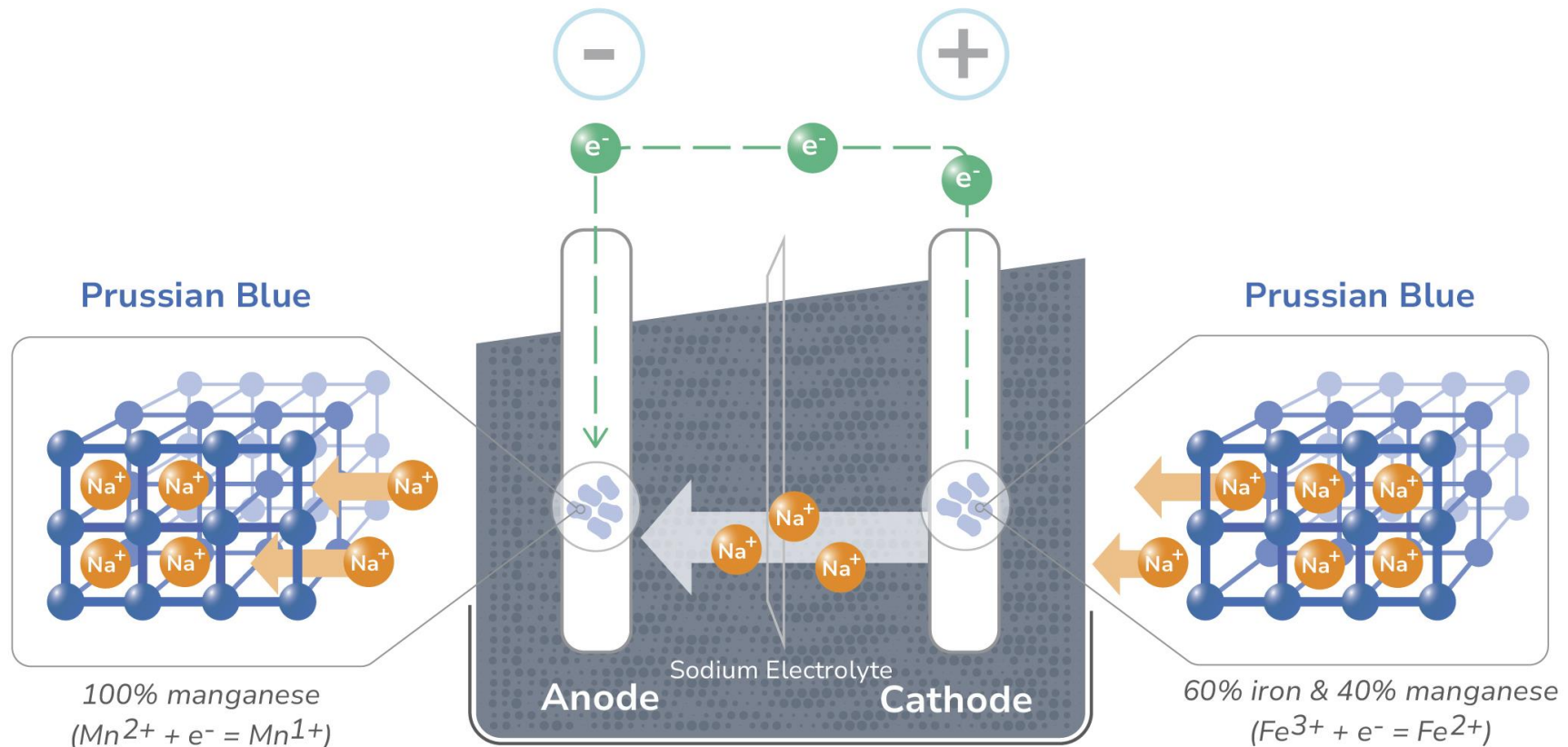


According to the International Air Transport Association, "...lithium-ion cells and batteries shipped by themselves must be shipped at a state of charge not exceeding 30% of their rated capacity. Lithium batteries are dangerous goods, and all of the regulatory requirements must be complied with..."



NATRON ENERGY

# Technology



**Natron's Sodium Ion Cell**  
Charging Phase  
(Discharge Phase - The flow would reverse)

## **Natron delivers more power, more often, and faster than any other chemistry**

- Half the internal resistance of lead acid
- Significantly greater percentage of total energy delivered during rapid discharge than other battery chemistries
- 70% of rated energy is delivered during 2-minute discharge
- 33% of rated energy is delivered during 30 second discharge
- Extremely rapid recharge with no settling required, and no cooling required
  - 0-99% SOC in <15 minutes
  - 0-70% SOC during 16C recharge lasting 2.5 minutes
  - 70-99% SOC during constant voltage hold lasting 6 minutes



# Environmental, Social, and Governance

**A Blue Battery for a Green Planet**

## Natron eliminates the “blood” minerals

- No lithium, cobalt, nickel, copper, or zinc
- Check others’ SDS declarations!

## Natron eliminates lead, a global public health crisis

- Uncontrolled emissions from lead smelting and recycling



Cobalt mining, Congo



Lead acid recycling, Indonesia



# Supply Chain



# Qualified for BABA and other Buy American Acts



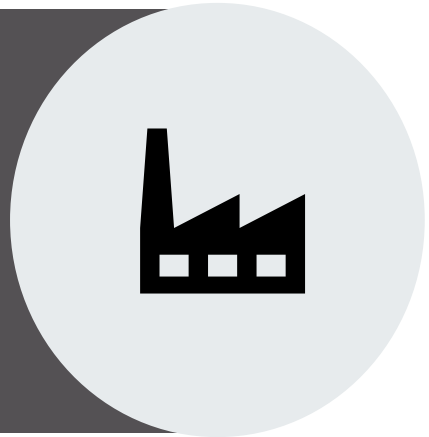
Michigan Production Facility



# Holland, Michigan Production Facility Details

The Holland plant will allow us to establish markets and secure giga-scale demand before we build our first gigafactory.

- **Plant has already proven the scalability of our process technology**
  - Materials: >1,000,000x scale-up complete
  - Current 500-ton/yr. capacity ready to scale to >10 kton
  - Electrodes: coating ~11.5km per shift
  - Cell assembly: first customer deliveries – December 2024
  - **Giga-factory (NC) will leverage the Holland plant's performance to operate multiple parallel production lines**



Automated Cell Stacking



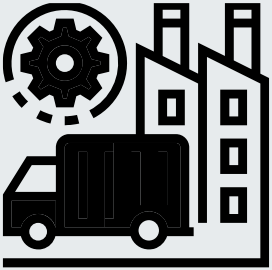
Tab Welding



Cell Testing Towers

# No Supply Chain Constraints!

- **Natron does not depend on questionable supply chains**
- **Natron can source from multiple North American supply chains if ever needed**



**Secure  
Supply Chain**

These are not  
an issue for Natron!



Geopolitical events



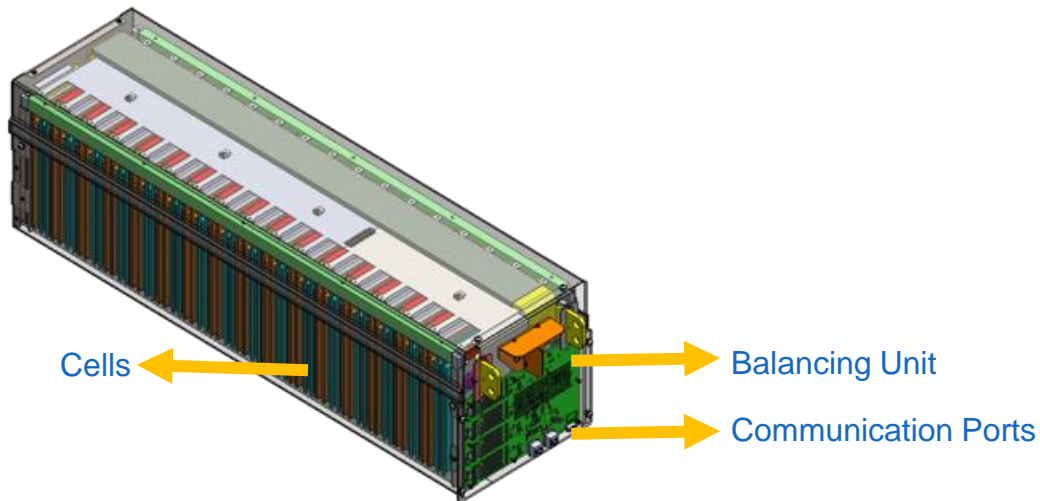
Delays at Ports



# Products

# BluePack™ 25kW, 48VDC Battery

- **Modular 25kW, 48VDC packs can be serialized for power systems from 96 to > 1,000VDC**
- **Operating Voltage range of 58 to 32 volts for max power/energy delivery**
- **800 Amp discharge & charge capable**
- **Class-leading charge and discharge**
  - 2-5 minute optimal discharge
  - Recharges in 10 minutes or less



Watch it being tested here: [https://www.youtube.com/watch?v=a3GURWZs\\_ec](https://www.youtube.com/watch?v=a3GURWZs_ec)

# BlueRack™ Battery Cabinets Shipping December 2024

- **Proof of Concept systems available now**
- **High Peak Power capacity eliminates need for N+1**
- **Higher power cabinets enable 2+ MVA UPS power blocks**
  - Fewer strings
  - Higher per cabinet standard power
  - Significantly higher Peak Power capacity
- **250 kW per cabinet nominal at a 2-minute discharge**
- **340 kW+ peak at <1 minute discharge rating**
- **Can be combined to make larger systems**
- **Other voltages available > 1,000 VDC**





# Applications And Use Cases

# Uninterruptible Power Supplies

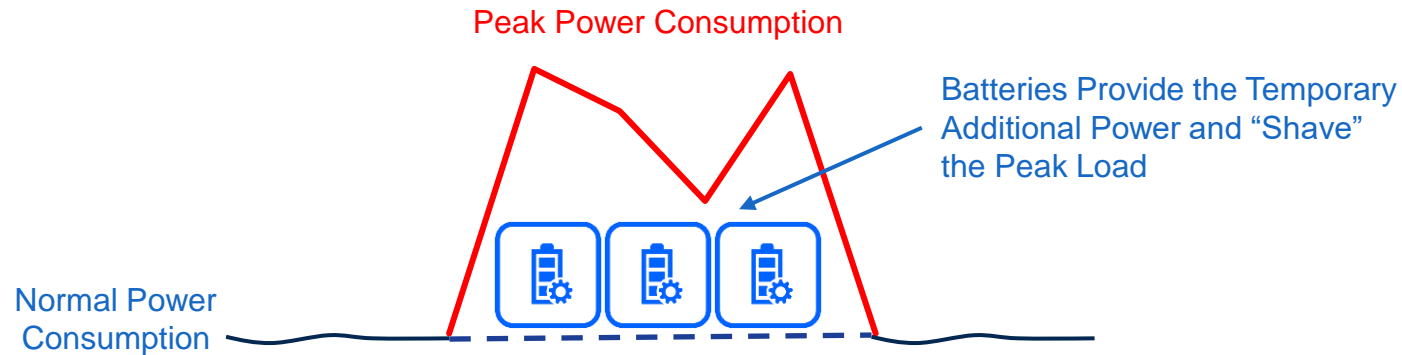


Natron Customer Demonstration Center





# Decarbonization – Displacing GenSets

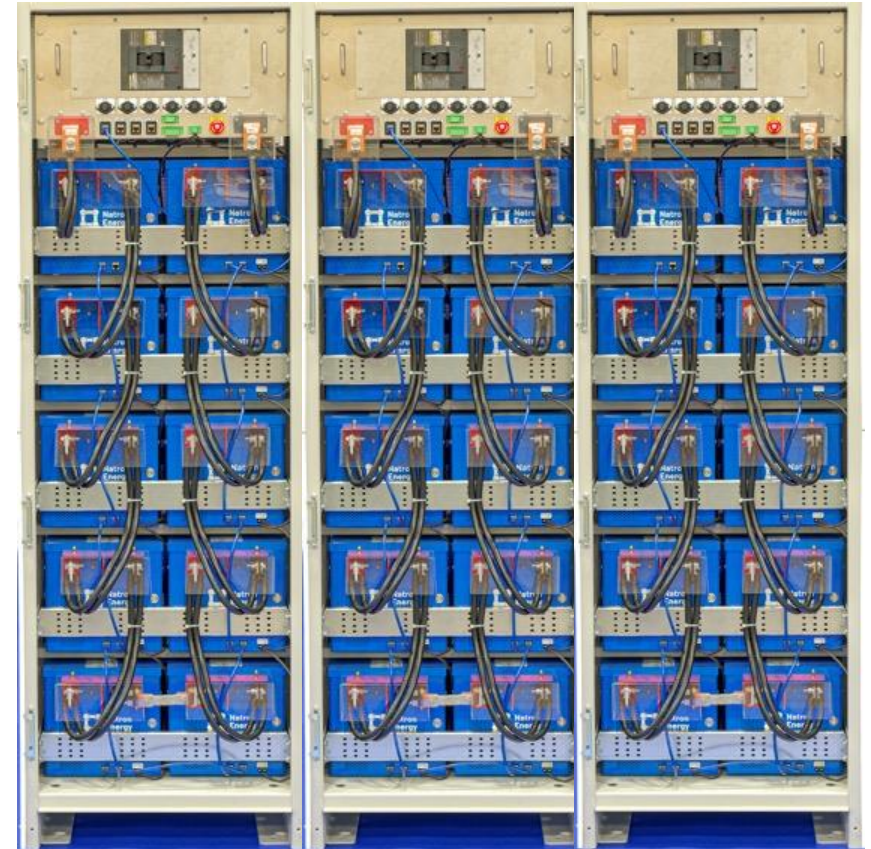
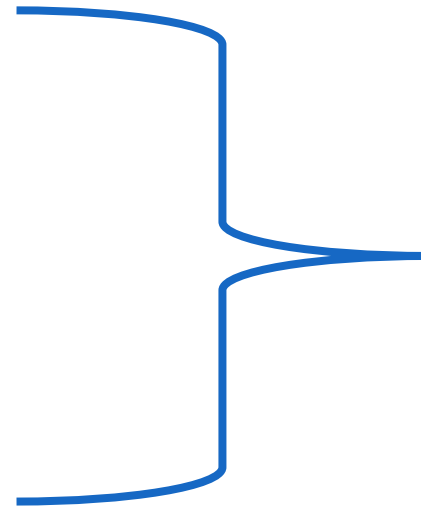
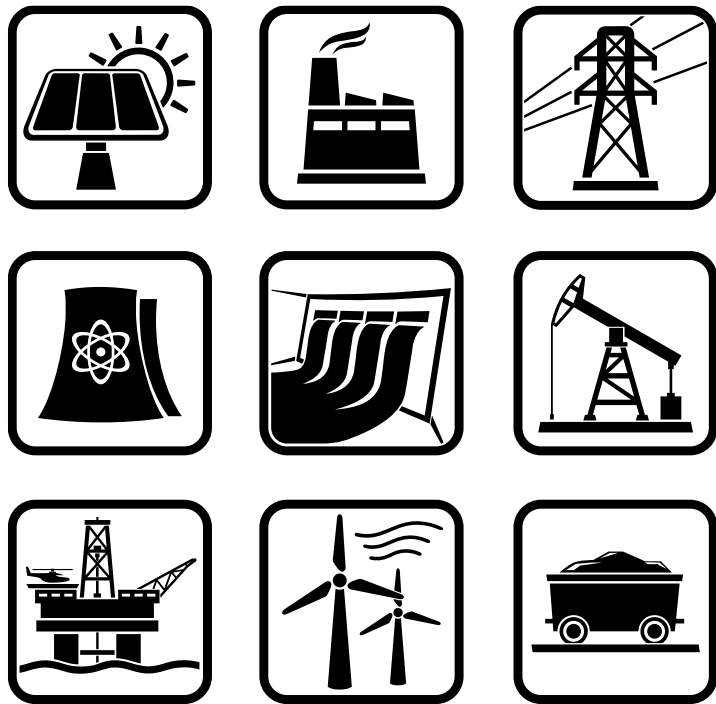


## Example

- Batteries used as peak shaving for drill rig power system, displacing a generator
- Diesel fuel savings business model
- Hundreds of sites globally
- Tried other storage systems first - FAILED
- Managing peak loads from seconds to 15 minutes
- Natron solution requires no maintenance
- Cycle life very appealing – can be hundreds of partial discharges per day
- Best TCO & lowest OPEX battery system available



# Micro-Grids & Grid Stabilization



## High Power for Engine & Turbine Starting systems

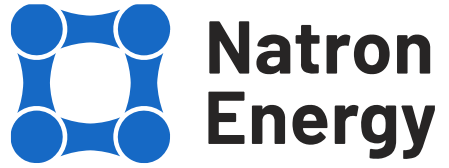


## Other Sodium Chemistries / Batteries

Each with their own unique operating parameters

- Faradion, UK, India – sodium-ceramic, safe, low peak power, long duration energy
- Horien, CH, former FZSoNick, high-temp SoNiCl long duration
- Tiamat, FR, – EV, cylindrical potential thermal runaway
- Northvolt, SW – energy storage, low cycle-count, low peak power
- China – CATL, BYD, HiNa – all EV-type, carbon anode, low peak power
  - Potential thermal runaway





Thank You



*Powered by Prussian Blue*